Phytochemical and larvicidal studies on Stemona curtisii: Structure of a new pyriodo[1,2-a]azepine Stemona alkaloid

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Publication Details

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Phytochemical and Larvicidal Studies on *Stemona curtisii*: Structure of a new Pyrido[1,2-a]azepine *Stemona* Alkaloid

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A new pentacyclic *Stemona* alkaloid, stemocurtisinol (3), with a pyrido[1,2-α]azepine A,B-ring system, and the known pyrrolo[1,2-α]azepine alkaloid oxyprotostemonine (4) have been isolated from a root extract of *S. curtisii*. The structure and relative stereochemistry of stemocurtisinol was determined by spectral data interpretation and X-ray crystallography. This compound is a diastereoisomer of oxystemokerrin and has the opposite configuration at C-4 and C-19. The individual alkaloid components showed significant larvicidal activity (IC₅₀ 4-39 ppm) on mosquito larvae (*Anopheles minimus* HO).
The *Stemona* group of alkaloids includes more than forty different natural products that have been structurally classified into five different groups. The pyrrolo[1,2-\(a\)]azepine (5,7-bicyclic A,B-ring system) nucleus is common to all compounds in these groups. In 2003 we reported the structure of stemocurtisine (1), the first example of a *Stemona* alkaloid with a pyrido[1,2-\(a\)]azepine A,B-ring system (that is, a 6,7-bicyclic A,B-ring system), isolated from the roots of *S. curtisii* Hook. Later in that year Hofer and Greger reported the isolation of five *Stemona* alkaloids with the pyrido[1,2-\(a\)]azepine A,B-ring system, including stemocurtisine (1), which they named pyridostemin, and oxystemokerrin (2) from an unidentified *Stemona* species (HG 915) and *S. kerrii*, respectively. Interestingly, these workers also examined the phytochemicals from *S. curtisii*; the major and minor components were stemofoline, a known pyrrolo[1,2-\(a\)]azepine *Stemona* alkaloid, and 2′-hydroxystemofoline, respectively, with trace amounts of oxystemokerrin (2) also detected. Our root samples of *S. curtisii* were collected in Trang Province in Southern Thailand in May 2002, and a voucher specimen was deposited in the Herbarium at Chiang Mai University. In turn the *S. curtisii* plant samples of Hofer and Greger were collected from Sutun Province in Southern Thailand (date not disclosed). Unfortunately, no voucher plant specimen from the collection of Hofer and Greger was deposited in Thailand, making a botanical comparison difficult. However, we are confident that Mr. J. Maxwell, a taxonomist from the Herbarium at Chiang Mai University and an expert of tropical plants, has identified the correct plant species by comparing our specimen with the key to the genera (*Stemona*) of Duyfjes and Prain. We report here the isolation of two further *Stemona* alkaloids.
from the roots of this plant and the larvicidal activity of the crude root extract of *S. curtisii*, and its three main alkaloid components, on mosquito larvae (*Anopheles minimus* HO).

Preparative TLC of the crude ethanol extracts of the dried root (400 g) of *S. curtisii* gave, stemocurtisine (1, 20.8 mg), stemocurtisinol (3, 12.7 mg) and oxyprotostemonine (4, 10.5 mg). The latter alkaloid was isolated in trace amounts from the roots of *S. kerrii* and *S. curtisii* by Hofer and Greger and the 1H and 13C NMR and MS data of 4 were almost identical to that reported by these workers.

Compound 3 was obtained as colorless prismatic crystals (mp 209-211 °C) by careful and slow evaporation of a solution of 3 in ethyl acetate. HRMS (EI +ve, m/z [M+] 405.2100, calcd 405.2151) indicated that 3 had the molecular formula C22H31NO6. The 1H and 13C NMR spectra of 3 indicated the presence of the ABCD-ring system of stemocurtisine (1), including the ether bridging structure between C-1 and C-9 and the C-4, 1-hydroxypropyl A-ring side chain. NOESY experiments showed significant cross peaks between H-19 and H-10α and H-19 and H-2β, revealing the β-configuration (axial orientation) of the C-4 substituent relative to the axial protons H-2β and H-10α (Figure 1). X-ray structural analysis confirmed the molecular formula of 3 and revealed its connectivity and relative stereochemistry (Figure 1). In the crystal, the hydroxyl hydrogen, H(19O), is hydrogen-bonded intramolecularly to N(5) at a distance of 2.04(3) Å (Figure 1). The absolute configuration of 3 was not established but is assumed, based on the known configurations of *Stemona* alkaloids with similar C,D-ring structures. The NMR data of 2 and 3 were significantly different, especially in their 13C NMR chemical shifts for the carbons near the C-4, 1-hydroxypropyl side chain, consistent with
these compounds being epimeric at C-4. Significant differences were observed in the chemical shifts for C-6 (δ 42.8 in 23 and δ 54.8 in 3) and C-19 (δ 70.5 in 23 and δ 67.9 in 3). The differences in the C-6 chemical shifts are consistent with the C-4 substituent having an equatorial disposition in 2 and an axial disposition in 3. Indeed the 13C NMR chemical shift for C-6 in 12 and 3 were almost identical, whereas that in 2 is about 12 ppm upfield due to the γ-gauche effect8 of the C-4 substituent on C-6. Compounds 2 and 3 also have opposite configurations at the carbinol carbon C-19.

The full 1H and 13C NMR spectral assignments for 3, based on extensive COSY, TOCSY, NOESY, HMQC, and HMBC experiments, are shown in Table 1. As described earlier, NOESY experiments were used to determine the relative α or β orientation of the protons.2

The larvicidal activity of the crude root extract of S. curtisii and that of compounds 1, 3 and 4 on mosquito larvae (Anopheles minimus HO), using the WHO method to determine the LC50,9 are shown in Table 2. While the crude ethanol extract showed a LC50 of 81 ppm, the individual alkaloid components were significantly more potent (LC50 of 4-39 ppm). The most potent was oxyprotostemonine 4 having a LC50 of 4 ppm.

In conclusion, two further pentacyclic Stemona alkaloids, stemocurtisinol (3) and oxyprotostemonine (4) have been isolated from a root extract of S. curtisii. Stemocurtisinol (3) is a diastereoisomer of oxystemokerrin (2) and has the opposite configuration at C-4 and C-19. The individual alkaloid components showed significant larvicidal activity on mosquito larvae (Anopheles minimus HO). The differences between the phytochemical profiles of S. curtisii studied by us and by Hofer and Gerber may be
due in part to the fact that these plants were harvested from different geographic regions and possibly at different seasons.\textsuperscript{10}

**Experimental Section**

**General experimental procedures.** As described previously.\textsuperscript{2}

**Plant Material.** The plant material [a voucher specimen is deposited at the Herbarium (number 17581) of the Department of Biology, Chiang Mai University] was collected as described previously.\textsuperscript{2}

**Extraction and Isolation.** The crude alkaloid material (0.9 g) that was described previously\textsuperscript{2} was chromatographed on silica gel (100 mL) using gradient elution from 100\% dichloromethane to 50\% methanol/dichloromethane containing 1\% concentrated aqueous ammonia as eluent. A total of 1500 mL of eluent was collected in test tubes of 25 mL. On the basis of TLC analysis these fractions were pooled to give 18 fractions. Fractions 12 (62.2 mg) and 13 (86.2 mg) were combined and re-chromatographed by preparative TLC (dichloromethane-methanol-aqueous ammonia, 94:5:1) to give 20.8 mg of pure stemocurtisine (1). Fraction 10 (136.8 mg) was chromatographed on silica gel using gradient elution from 100\% dichloromethane to 50\% methanol/dichloromethane containing 1\% concentrated aqueous ammonia as eluent, six fractions were collected. Fraction 2 (65.8 mg) and fraction 3 (30.6 mg) were re-chromatographed by preparative TLC (dichloromethane-methanol-ammonia, 96:4:1) to give 12.7 mg of pure stemocurtisinol (3) and 10.5 mg of pure oxyprotostemonine (4), respectively. The $^1$H and $^{13}$C NMR data of 4 were identical to that reported in the literature.\textsuperscript{3}
Stemocurtisinol (3): pale yellow needles (ethyl acetate); mp 209-211 °C; \([\alpha]_{D}^{25} +233^\circ\ (c\ 0.334,\ \text{CHCl}_3);\ \text{IR}\ (\text{film})\ \nu_{\text{max}}\ 2930,\ 1746,\ 1692,\ 1621,\ 1461,\ 1397,\ 1367,\ 1286,\ 1216,\ 1155,\ 1025,\ 994.62,\ 754\ \text{cm}^{-1};\ ^1\text{H}-\ \text{and}\ ^{13}\text{C NMR},\ \text{see Table 1};\ \text{HREIMS}\ \text{m/z 405.2100}\ [\text{M}]^+,\ \text{calcd for C}_{22}\text{H}_{31}\text{NO}_6\ 405.2151.

X-ray Structure Determination of 3: \(\text{C}_{22}\text{H}_{31}\text{NO}_6,\ M = 405.5.\ \text{Orthorhombic,}\ \text{space group}\ P2_1\text{2}_1\text{2}_1\ (D^4_2,\ \text{No.}\ 19),\ a = 7.708(3),\ b = 14.177(6),\ c = 18.972(8)\ \text{Å,}\ V = 2073\ \text{Å}^3.\ \text{Dc}\ (Z = 4) = 1.299\ \text{g cm}^{-3}.\ \mu_{\text{Mo}} = 0.09\ \text{mm}^{-1};\ \text{specimen:}\ 0.45\times0.15\times0.13\ \text{mm; }\ T_{\text{min}}/\text{max}\ (\text{multiscan correction}) = 0.90.\ 2\theta_{\text{max}} = 58^\circ;\ N_{\text{total}}\ (\text{full sphere}) = 26220,\ \text{merging to } N(\text{unique}) = 3015\ (R_{\text{int}} = 0.035),\ N_{\text{obs}}\ (F>4\sigma(F)) = 2631;\ R = 0.035,\ R_w = 0.041\ (\text{weights: } (\sigma^2(F) + 0.0003F^2)^{-1}).\ \langle x,y,z,U_{iso}\rangle\ H\ \text{refined.}\ |\Delta\rho_{\text{max}}| = 0.22(2)\ \text{e Å}^{-3}.\ \text{Bruker AXS instrument, } T \approx 153\ \text{K; monochromatic Mo K}\alpha\ \text{radiation,}\ \lambda = 0.71073\ \text{Å. CCDC # 226011.}

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References and Notes


(5) Prain, D. J. Asiatic Soc. Bengal, 1905, 73, 39-44.


1, R = H

2, R =

3

4
Figure 1. Molecular projection of 3, showing 50% probability amplitude displacement ellipsoids, hydrogen atoms having arbitrary radii of 0.1 Å.
Table 1. $^{13}$C NMR (75 MHz) and $^1$H NMR (500 MHz) Spectral Data of 3 in CDCl$_3$ Solution.

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<th>position</th>
<th>$\delta$ C (DEPT)</th>
<th>$\delta$ H (mult., J (Hz), assign.)</th>
<th>HMBC</th>
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<tr>
<td>1</td>
<td>75.4 (CH)</td>
<td>4.05 (s) 4.05 (s)</td>
<td>H-2$\alpha$, H-2$\beta$, H-3$\alpha$, H-3$\beta$, H-10, H-10a</td>
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<td>2</td>
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<td>1.73 (dd, 5.8, 12.3, $\beta$)</td>
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<td>18.4 (CH$_2$)</td>
<td>1.36 (m, $\beta$) 1.96 (m, $\alpha$)</td>
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<td>65.5 (CH)</td>
<td>2.53 (m)</td>
<td>H-3$\alpha$, H-3$\beta$, H-6$\alpha$, H-6$\beta$, H-19</td>
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<td>6</td>
<td>54.8 (CH$_2$)</td>
<td>2.92 (dd, 4.5, 15.5, $\alpha$)</td>
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<td>25.8 (CH$_2$)</td>
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<td>8</td>
<td>33.5 (CH$_2$)</td>
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<td>9</td>
<td>120.1 (C)</td>
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**Table 2.** Larvicidal Activity of *S. curtisii* on Mosquito Larvae (*Anopheles minimus* HO) Using the WHO Method.$^9$