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Effects of salinity build-up on the performance of an anaerobic membrane bioreactor regarding basic water quality parameters and removal of trace organic contaminants

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Abstract

The effects of elevated inorganic salt concentration on anaerobic membrane bioreactor (AnMBR) treatment regarding basic biological performance and trace organic contaminant (TrOC) removal were investigated. A set of 33 TrOCs were selected to represent pharmaceuticals, steroids, and pesticides in municipal wastewater. Results show potential adverse effects of increase in the bioreactor salinity to 15 g/L (as NaCl) on the performance of AnMBR with respect to chemical oxygen demand removal, biogas production, and the removal of most hydrophilic TrOCs. Furthermore, a decrease in biomass production was observed as salinity in the bioreactor increased. The removal of most hydrophobic TrOCs was high and was not significantly affected by salinity build-up in the bioreactor. The accumulation of a few persistent TrOCs in the sludge phase was observed, but such accumulation did not vary significantly as salinity in the bioreactor increased.

Keywords

contaminants, performance, salinity, anaerobic, membrane, build, bioreactor, regarding, effects, basic, water, up, quality, parameters, removal, trace, organic

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27 **ABSTRACT**

28 The effects of elevated inorganic salt concentration on anaerobic membrane bioreactor
29 (AnMBR) treatment regarding basic biological performance and trace organic contaminant
30 (TrOC) removal were investigated. A set of 33 TrOCs were selected to represent
31 pharmaceuticals, steroid, pesticides in municipal wastewater. Results show potential adverse
32 effects of increasing in the bioreactor salinity to 15 g/L (as NaCl) on the performance of
33 AnMBR with the respect to the COD removal, biogas production, and the removal of most
34 hydrophilic TrOCs. Furthermore, a decrease in biomass production was observed as salinity
35 in the bioreactor increased. The removal of most hydrophobic TrOCs was high and was not
36 significantly affected by salinity build-up in the bioreactor. The accumulation of a few
37 persistent TrOCs in the sludge phase was observed, but such accumulation did not vary
38 significantly as salinity in the bioreactor increased.

39 **Key words:** Salinity build-up; anaerobic membrane bioreactor (AnMBR); trace organic
40 contaminants (TrOCs); wastewater treatment; biogas production.

41 **1 Introduction**

42 Water scarcity is a vexing challenge to the sustainable development of our society. This issue
43 is further exacerbated by climate change, continuous population growth, industrialization and
44 urbanization, and environmental pollution (Shannon et al., 2008). Moreover, an increasing
45 number of trace organic contaminants (TrOCs) – including pharmaceuticals and personal
46 products, endocrine disrupting compounds, and pesticides – are continuously released to the
47 aquatic environmental through sewage effluent discharge and other human activities. This
48 continuous release of TrOCs can compromise our limited water resources for drinking water
49 supply (Schwarzenbach et al., 2006). As a result, much attention has been dedicated to the
50 removal of TrOCs during wastewater treatment and to explore alternative water sources
51 including wastewater to protect and increase water supply.

52 Membrane bioreactor (MBR) is a promising technology for wastewater treatment and water
53 reuse (Judd et al., 2011; Hai et al., 2014; Jegatheesan et al., 2016). Recent studies have shown
54 that MBR can have higher removal of some TrOCs in comparison to conventional activated
55 sludge treatment (De Wever et al., 2007; Melvin et al., 2016). The observed enhanced TrOC
56 removal can be attributed to the prolonged solid retention time (SRT) and high biomass

57 concentration in the MBR systems (Hai et al., 2014). It is noteworthy that the removal of
58 TrOCs by MBR investigated in most of previous studies was under an aerobic condition.

59 MBR can also be deployed in anaerobic configuration (i.e. AnMBR) (Liao et al., 2006; Lew
60 et al., 2009; Skouteris et al., 2012). Compared to its aerobic counterpart, AnMBR is much
61 more energy efficient due to the absence of aeration and enables the treatment of high
62 strength wastewater with less sludge production (Skouteris et al., 2012). More importantly,
63 biogas can be produced for beneficial use during AnMBR treatment. As a result, AnMBR has
64 attracted much research interest over last decade and its industrial application is increasing
65 remarkably (Lin et al., 2013). Most AnMBR studies have focused on the treatment of high
66 strength industrial wastewater (Saddoud et al., 2009; Stamatelatou et al., 2009; Dereli et al.,
67 2012). Compared to industrial wastewater, municipal wastewater has much lower strength due
68 to its dilution nature. Thus, anaerobic treatment may not suit to treat municipal wastewater
69 given its long operating hydraulic retention time (HRT), energy requirement to maintain a
70 mesophilic digestion temperature (approximately 35 °C), and large wastewater volume (Lew
71 et al., 2009; Hai et al., 2014).

72 Recent interest to simultaneously recover energy and clean water during wastewater
73 treatment has spurred new research to adapt AnMBRs for municipal wastewater treatment.
74 One viable technique is to pre-concentrate the organic content (usually measured as chemical
75 oxygen demand (COD)) of municipal wastewater to a range suitable for anaerobic treatment
76 (Diamantis et al., 2013). This aim can be achieved by directly extracting clean water from
77 municipal wastewater using forward osmosis or other high-retention membrane processes,
78 resulting in a concentrated sewage solution (Xie et al., 2013; Zhang et al., 2014). However,
79 the pre-concentration process prior to AnMBRs also entails the build-up of salinity in the
80 concentrated municipal wastewater (Ansari et al., 2015). Moreover, since a high-retention
81 membrane process can effectively retain TrOCs (Luo et al., 2014), their concentrations in
82 pre-concentrated wastewater prior to AnMBR can be an order of magnitude higher than those
83 in the initial wastewater solution. In addition, varying salinity of municipal wastewater also
84 occurs in coastal regions due to seawater infiltration to sewers or when sewer systems receive
85 discharges from industrial processes that involve saline water, such as seafood and cheese
86 production (Yogalakshmi et al., 2010).

87 High salinity wastewater is a challenge to biological treatment (Lay et al., 2010). Elevated
88 salinity can negatively affect the performance of aerobic MBR by inhibiting microbial

89 activity and growth (Yogalakshmi et al., 2010). An increase in the osmotic stress can result in
90 the dehydration and plasmolysis of microbial cells and thus their inactivity (Wood, 2015).
91 Nevertheless, microbial acclimatization can lead to the succession of halotolerant and even
92 halophilic bacteria, thereby gradually recovering the treatment performance (Luo et al.,
93 2016). However, compared to aerobic MBR, little is known about the effects of high salinity
94 on the performance of anaerobic MBR.

95 This study aims to investigate the effects of salinity build-up on the performance of AnMBR,
96 particularly in terms of TrOC removal. Salinity build-up was stimulated by increasing the
97 influent NaCl loading from 0 to 15 g/L. Basic performance of AnMBR was evaluated with
98 respect to bulk organic removal, biomass growth, and biogas/methane production. Removal
99 of TrOCs by AnMBR under the elevated salinity condition was related to their
100 physicochemical properties, such as hydrophobicity and molecular structure. Results in this
101 study would shed lights on the management of saline wastewater before AnMBR treatment.

102 **2 Materials and methods**

103 *2.1 Synthetic wastewater and trace organic contaminants*

104 A synthetic wastewater with approximately 6,000 mg/L COD (Table S1, Supplementary
105 Data) was used to simulate high strength municipal wastewater and to maintain stable
106 influent conditions. A concentrated stock solution was prepared every 5 days and kept at 4
107 °C. The synthetic wastewater was prepared daily by diluting the concentrated stock solution
108 with deionized water.

109 A set of 33 TrOCs, representing four key groups of emerging contaminants of significant
110 concerns that present ubiquitously in municipal wastewater (i.e. pharmaceuticals, personal
111 care products, industrial chemicals, and pesticides), were selected in this study. Key
112 properties - including hydrophobicity and molecular structure - of these TrOCs are
113 summarized in Table S2 of the Supplementary Data. These TrOCs can be classified as
114 hydrophobic or hydrophilic depending on their effective octanol-water partition coefficient
115 (denoted as Log D). Compounds with log D at solution pH 7 higher than 3.2 are hydrophobic
116 whereas compounds with log D at solution pH 7 lower than 3.2 are hydrophilic in a neutral
117 condition (Tadkaew et al., 2011). A stock solution containing all 33 TrOCs (10 mg/L of each)
118 was prepared in pure methanol and stored at -18 °C in the dark. The stock solution was used
119 within one month. Regular measurements were conducted to confirm the constant
120 concentration of the TrOC stock solution.

121 2.2 *Experimental system and protocol*

122 A lab-scale AnMBR system was used in this study (Figure S1, Supplementary Data). This
123 system comprised a 30 L stainless steel bioreactor, an external ceramic microfiltration (MF)
124 membrane module (NGK, Japan), and several peristaltic and circulation pumps. The MF
125 membrane had a pore size of 0.1 μm and an effective area of 0.09 m^2 . A PID regulated heater
126 (Neslab RTE7, Thermo Scientific, USA) equipped with a plastic heater exchange coil was
127 used to maintain the bioreactor temperature at 35 ± 1 $^{\circ}\text{C}$ over the entire experimental period.
128 A peristaltic pump (Masterflex L/s, USA) controlled by water level controller was used to
129 feed the bioreactor, which had a constant working volume of 20 L. The digested sludge was
130 circulated from the bioreactor to the external membrane module and then back to the
131 bioreactor by a peristaltic pump with a circulation rate of 700 mL/min. At the same time, an
132 industrial grade peristaltic hose pump (ProMinent, Australia) was used to mix the sludge by
133 circulating it from the bottom to the top of the bioreactor. A Tedlar sampling bag was
134 connected to the bioreactor for biogas collection. Both the bioreactor and pipes involved in
135 this system were rapped with insulation foam to reduce heat loss. A detailed description of
136 this system is also available elsewhere (Wijekoon et al., 2015).

137 Anaerobic sludge collected from the Wollongong Wastewater Treatment Plant was used to
138 inoculate the bioreactor with feeding the synthetic wastewater described above for over 12
139 months. Once acclimatized in term of bulk organic removal (i.e. COD removal > 96%),
140 TrOCs were spiked to the synthetic wastewater on a daily basis to obtain a working
141 concentration of 2 $\mu\text{g/L}$ of each compound. The initial mixed liquor suspended solids (MLSS)
142 concentration was adjusted to approximately 16 g/L. Salinity build-up in the bioreactor was
143 induced by increasing the influent NaCl loading from 0 to 15 g/L with an increase of 1 g/L
144 per day (Figure S2, Supplementary Data). To allow microbial acclimatization to the salinity
145 stress, the influent salt salinity was maintained at 5, 10, and 15 g/L NaCl for two weeks. The
146 MF membrane was operated in a cycle of 14 min suction and 1 min relaxation with a water
147 flux of 1.8 $\text{L/m}^2\text{h}$, which resulted in an operating HRT of 5 days. The low water flux and
148 relaxation time was provided to reduce membrane fouling. No sludge was wasted in this
149 study, except for regular sludge sampling, which led to an operating SRT of 140 days.
150 Sodium acetate was added to maintain the bioreactor pH of 7. The MF membrane was
151 chemically cleaned once a month by using a 20 mg/L NaOH solution at 70 ± 1 $^{\circ}\text{C}$ and then
152 completely rinsed with deionized water. This cleaning procedure could completely recover

153 the membrane permeability determined by the measured transmembrane pressure and water
154 flux with deionized water as the feed.

155 2.3 *Analytical methods*

156 2.3.1 *Basic measurements*

157 MLSS and mixed liquor volatile suspended solids (MLVSS) concentrations were measured
158 according to the Standard Methods for Examination of Water and Wastewater (APHA, 2005).
159 Total organic carbon (TOC) and total nitrogen (TN) were analysed using a TOC/TN-VCSH
160 analyser (Shimadzu, Japan). COD was measured using high range plus digestion vials (Hatch,
161 USA) following the standard dichromate method. Mixed liquor electrical conductivity and
162 pH were monitored by an Orion 4 Star Plus portable pH/conductivity meter (Thermo
163 Scientific, USA). Biogas composition was revealed by a biogas meter (Biogas 5000, Geotech,
164 UK).

165 2.3.2 *TrOC analysis*

166 Aqueous samples (250 mL) were taken twice (once per week) from the feed and permeate
167 when the salinity was stabilized at 0, 5, 10, and 15 g/L NaCl to analyse TrOC concentrations
168 based on the method described previously by Tadkaew et al. (2011). Briefly, this method
169 involved solid phase extraction (SPE), liquid chromatography, and quantitative measurement
170 by tandem mass spectrometry with electrospray ionization. All samples were spiked with a
171 surrogate solution that contained 50 ng of each TrOC in an isotopically labelled version. The
172 use of isotope dilution allows for SPE efficiency correction and complete elimination of any
173 matrix effects (Trenholm et al., 2006). Oasis HLB cartridges (Waters, Millford, MA, USA)
174 used for TrOC extraction were preconditioned using 5 mL methyl tert-butyl ether, 5 mL
175 methanol, and 5 mL reagent water (two times). The cartridges were rinsed twice with 5 mL
176 reagent water after SPE and then processed for nitrogen drying.

177 TrOCs were eluted from the loaded cartridges using 5 mL methanol, and then 5 mL mixture
178 of methanol and methyl tert-butyl ether (1:9, v/v). Resultant extracts were concentrated to
179 100 μ L by using nitrogen stream, which were subsequently diluted to 1 mL with methanol.
180 The diluted extracts were processed to a high performance liquid chromatography (Agilent
181 1200 series, Palo Alto, CA, USA) with a Luna C18 (2) column (Phenomenex, Torrance CA,
182 USA) for TrOC separation. Peaks of different TrOCs were identified and quantified by an
183 isotope dilution method using a triple quadrupole mass spectrometer (API 4000, Applied
184 Biosystems, Foster City, CA, USA) equipped with a turbo-V ion source that was employed in

185 both positive and negative electro-spray modes. This measurement method had a limit of
186 quantification of 20 ng/L for bisphenol A, 10 ng/L for caffeine, triclocarban and diuron, and 5
187 ng/L for all other TrOCs.

188 The removal of TrOCs by the AnMBR system was determined from:

$$189 \quad R = \frac{C_f - C_p}{C_f} \times 100\%$$

190 where C_f and C_p were the measured TrOC concentrations in the feed and permeate,
191 respectively.

192 TrOCs resided in the sludge were measured twice (once per week) when the salinity was
193 stabilized at 0, 5, 10, and 15 g/L NaCl based on a method previously reported by Wijekoon et
194 al. (2013). In brief, the mixed liquor was centrifuged at 3750g for 20 mins to obtain sludge
195 pellet, which was then freeze-dried using a Freeze Dryer (Alpha 1–2 LDplus, Christ GmbH,
196 Germany). The dried sludge was completely ground and 0.5 g sludge powder was mixed with
197 5 mL methanol in a glass valve using a vortex mixer (VM1, Ratek, Australia). The mixture
198 was ultrasonicated at 40 °C for 10 min and then centrifuged (3270g for 10 min). The
199 supernatant was collected while the remaining pellet was mixed with 5 mL dichloromethane
200 and methanol mixture (1:1, v/v), and then processed for ultrasonication and centrifugation.
201 Supernatant collected from these two steps was purged with nitrogen gas to removed residual
202 methanol and dichloromethane, and then diluted to 250 mL with Milli-Q water for TrOC
203 analysis using the method described above for aqueous samples.

204 **3 Results and discussion**

205 *3.1 Basic performance*

206 *3.1.1 Removal of bulk organic matter*

207 Small and transient decrease in the TOC removal by AnMBR was observed as the the
208 bioreactor salinity increased (Figure 1). At baseline condition (i.e. negligible salinity in the
209 bioreactor), the TOC removal was constant at approximately 98%. When salinity in the
210 bioreactor increased to 5 g/L NaCl, the TOC removal decreased to 82%. This observed
211 decrease was temporary and could be attributed to the negative effect of the elevated
212 bioreactor salinity on the digester activity. It has been reported that salinity increase could
213 resulted in cell plasmolysis and the loss of metabolic activity either in anaerobic or aerobic
214 conditions (Lay et al., 2010). Similar to that in aerobic MBR systems, microbial

215 acclimatization to the saline condition recovered the TOC removal to the initial level (i.e. 98%
216 removal). No significant impact on the TOC removal was observed even when the bioreactor
217 salinity continuously increased up to 15 g/L NaCl.

218 **[FIGURE 1]**

219 The elevated bioreactor salinity reduced the COD removal by AnMBR, particularly at the
220 salinity above 10 g/L NaCl (Figure 1). Similar to the TOC removal, at baseline condition (i.e.
221 negligible salinity in the bioreactor), the COD removal was more than 98%. There was no
222 notable effect on the COD removal as the bioreactor salinity increased to less than 10 g/L
223 NaCl. This observation is in good agreement with that reported by Gu et al. (2015) who
224 reported that the biological COD removal was relatively stable although the mixed liquor
225 electrical conductivity increased up to 20 mS/cm (corresponding to approximately 10 g/L
226 NaCl) during the operation of an anaerobic osmotic membrane bioreactor (AnOMBR) at a
227 mesophilic condition. However, a dramatic decrease in the COD removal (to approximately
228 80%) was observed when the bioreactor salinity rose beyond 10 g/L NaCl (Figure 1).
229 Previous studies have also reported the negative impact of such high salinity on the COD
230 removal by anaerobic processes, such as upflow anaerobic sludge blanket reactor (Aslan et al.,
231 2016) and sequential anaerobic and aerobic treatment (Shi et al., 2014). Although there was
232 some evidence of treatment recovery possibly due to microbial acclimatization, the
233 downward trend of COD removal under highly saline conditions (i.e. salinity >10 g/L NaCl)
234 persisted. These results suggest that salinity build-up in the bioreactor beyond 10 g/L NaCl
235 could adversely affect the AnMBR performance.

236 Results in Figure 1 show that AnMBR exhibited different variations in the removal of TOC
237 and COD in response to the salinity increase. This difference was possibly due to the
238 susceptibility of microbial communities (that were responsible for the biodegradation of un-
239 oxidisable organic matter) to the low saline stress. Nevertheless, further studies are necessary
240 to track changes in microbial community structure in response to the elevated bioreactor
241 salinity during AnMBR treatment.

242 Without a nitrification step, TN removal by anaerobic digesters is limited and mainly relies
243 on microbial assimilation. In this study, a significant decrease in the TN removal was
244 observed at the beginning of AnMBR operation without NaCl addition (Figure 1). The reason
245 for such decrease is not clear, but was probably due to the adverse impacts of methanol (used

246 to dissolve TrOCs) on nitrogen assimilation by digesters. As the bioreactor salinity gradually
247 increased up to 15 g/L NaCl, the TN removal only fluctuated in the range of 10 – 20%.

248 3.1.2 *Biogas production*

249 Biogas production was relatively stable (0.4 – 0.6 L/g COD_{loaded}) in response to an increase in
250 bioreactor salinity during AnMBR operation (Figure 2). Only a small decrease was observed
251 as the salinity increased to above 10 g/L NaCl. This observation is consistent with the
252 decreased COD removal at such high salinity (Figure 1). Nevertheless, the methane
253 composition in the produced biogas was stable in the range of 58 – 65% over the entire
254 experimental period (Figure 2), which is similar to that reported in a recent study (Wijekoon
255 et al., 2015), where the AnMBR system was operated for over 140 days under the same
256 conditions but without loading NaCl in the feed. These results indicate that salinity build-up
257 in bioreactor (up to 15 g/L NaCl) may not significantly affect the bioactivity of
258 methanogenesis. Gu et al. (2015) also observed a stable methane yield regardless of salinity
259 build-up in the bioreactor during AnOMBR operation.

260 [FIGURE 2]

261 3.1.3 *Biomass concentration*

262 Salinity build-up in the bioreactor reduced the active digesters during AnMBR operation
263 (Figure 3). At the baseline condition (i.e. negligible salinity in the bioreactor), both MLSS
264 and MLVSS concentration were relatively stable with the MLVSS/MLSS ratio at
265 approximately 0.7, suggesting that most digesters in the mixed liquor were active. As the
266 bioreactor salinity was enhanced to higher than 10 g/L NaCl, an increase in the MLSS
267 concentration (from 16 to 22 g/L) was observed while the MLVSS concentration decreased
268 significantly. This observation could be attributed to the negative effects on the bioactivity of
269 anaerobic digesters. Similar results have also been reported in aerobic MBR systems, in
270 which the elevated salinity resulted in dead cells and increased the secretion of extracellular
271 polymeric substances in the bioreactor, thus increasing the MLSS but reducing the MLVSS
272 concentrations (Tadkaew et al., 2013; Luo et al., 2015).

273 [FIGURE 3]

274 3.2 *Removal of trace organic contaminants*

275 A qualitative framework has been previously developed and evaluated by Wijekoon et al.
276 (2015) to predict the removal of various TrOCs by AnMBR based on their physicochemical

277 properties, mainly including hydrophobicity and molecular structure. A similar predictive
278 framework has also been widely applied to evaluate TrOC removal by aerobic MBR
279 (Tadkaew et al., 2011). As noted in Section 2.1, the 33 TrOCs selected in current study could
280 be classified as hydrophobic (i.e. $\text{Log } D > 3.2$) and hydrophilic (i.e. $\text{Log } D < 3.2$). Therefore,
281 the removal of TrOCs by AnMBR under the elevated bioreactor salinity was related to their
282 physicochemical properties based on these predictive frameworks (Figure 4).

283 **[FIGURE 4]**

284 *3.2.1 Removal of hydrophobic trace organic contaminants*

285 The removal of hydrophobic TrOCs (with $\text{Log } D > 3.2$ at pH 7) by AnMBR was higher than
286 80% with a few exceptions (including phenylphenol, bisphenol A, and triclosan) (Figure 4a).
287 More importantly, despite the decreasing active digester concentration (Figure 3), the
288 removal of most of these hydrophobic TrOCs was not significantly affected by the elevated
289 bioreactor salinity. The high removal of these compounds could be attributed to their
290 effective adsorption onto sludge, which could increase their biodegradation (Monsalvo et al.,
291 2014; Wijekoon et al., 2015).

292 Relatively low removal rates were observed for three hydrophobic compounds, including
293 phenylphenol, bisphenol A, and triclosan (Figure 4a). The removal of phenylphenol was only
294 60% at baseline salinity (i.e. no NaCl addition) and decreased at the bioreactor salinity higher
295 than 10 g/L NaCl. Such low removal could be due to the relatively low hydrophobicity of
296 phenylphenol ($\text{Log } D = 3.3$ at pH7). By contrast, the removal of clozapine (which had a
297 lower hydrophobicity than phenylphenol) was in the range of 80 – 98% although a small
298 decrease was observed with salinity increase. The observed difference in the removal of these
299 two compounds likely results from their different biodegradability, which determines the
300 mineralization of TrOCs in biological treatment. Bisphenol A was poorly removed and its
301 removal rate reduced from 40 to 20% as the bioreactor salinity climbed from negligible to 15
302 g/L NaCl. The low removal of bisphenol A is consistent with that reported by Monsalvo et al.
303 (2014) and could be ascribed to its low adsorption onto digesters although it had a relative
304 high hydrophobicity ($\text{Log } D = 3.6$ at pH 7). On the other hand, the removal of triclosan
305 increased from 40 to 60% with salinity increase up to 15 g/L NaCl. This result was possibly
306 due to the enhanced adsorption of triclosan on the digesters as salinity increased (Figure 5a).

307 3.2.2 Removal of hydrophilic trace organic contaminants

308 The removal of hydrophilic TrOCs ($\text{Log } D < 3.2$ at pH 7) varied significantly during AnMBR
309 operation at baseline salinity (i.e. negligible salinity in the bioreactor) (Figure 4b). This result
310 is in good agreement with that reported by Wijekoon et al. (2015) who attributed such
311 varying removal to the different biodegradability of these hydrophilic TrOCs, which was
312 further determined by their molecular structures. Similar results have also been reported in
313 anaerobic MBR treatment (Tadkaew et al., 2011). In this study, several hydrophilic TrOCs,
314 including trimethoprim, carazolol, hydroxyzine, amitriptyline, and linuron, were highly
315 removed (with removal rates above 80%). Such effective removal was due to their high
316 biodegradability with presence of electron donating functional groups, such as hydroxyl and
317 amine, in the molecular structure (Table S2, Supplementary Data). On the other hand, relative
318 low removal rates were observed for other hydrophilic TrOCs due to their resistance of
319 anaerobic biodegradation with the presence of electro withdrawing groups (e.g. chlorine and
320 amide) in their molecular structures (Wijekoon et al., 2015).

321 The elevated bioreactor salinity significantly reduced the removal of most hydrophilic TrOCs
322 (Figure 4b). Similar results have also been reported by Luo et al. (2015) although an aerobic
323 MBR with activated sludge was used in their study. These results suggest that the inhibition
324 of sludge metabolic activity caused by salinity build-up in the bioreactor could adversely
325 affect the removal of hydrophilic TrOCs either under aerobic or anaerobic conditions.
326 Nevertheless, a decrease but subsequent increase in the removal rate was observed for
327 trimethoprim. This observation could be attributed to the acclimatization of microbial species
328 that were responsible for trimethoprim biodegradation to the saline stress.

329 Of the 24 hydrophilic TrOCs investigated in this study, the removal of three compounds (i.e.
330 verapamil, hydroxyzine, and simazine) increased with salinity build-up in the bioreactor. The
331 enhanced removal of verapamil and hydroxyzine could be attributed to an increase in their
332 adsorption onto sludge as the bioreactor salinity elevated (Figure 5b). By contrast, the
333 adsorption of simazine was constantly negligible over the entire experimental period.
334 Therefore, the increased overall removal of simazine by AnMBR was possibly due to the
335 development of salt-tolerant bacteria that specifically target the compound. Nevertheless,
336 future studies are needed to relate such removal behaviours to the variation of microbial
337 community structure in response to the elevated bioreactor salinity.

338 3.2.3 Adsorption of trace organic contaminants onto sludge

339 Hydrophobicity and biodegradability of TrOCs are important factors determining their
340 residuals in the sludge. In this study, the accumulation of hydrophobic TrOCs was relatively
341 low in the digesters, although they were supposed to highly adsorb onto sludge (Figure 5a).
342 This observation could be attributed to the readily biodegradable nature of these compounds.
343 A fluctuated but discernable increase in the residual content was observed for several
344 compounds in response to the elevated bioreactor salinity. These compounds included
345 clozapine, bisphenol A, triclosan, triclocarban, and nonylphenol. Of the five compounds, the
346 increased accumulation in the sludge was more significant for clozapine and bisphenol A,
347 possibly due to their disrupted biodegradation at high salinity (Figure 4a). On the other hand,
348 the digesters might be more hydrophobic at high salinity condition, thereby enhancing the
349 adsorption of triclosan, triclocarban, and nonylphenol, which were highly hydrophobic.

350 [FIGURE 5]

351 No significant accumulation in the sludge was observed for hydrophilic TrOCs, with a few
352 exceptions, including carazolol, verapamil, hydroxyzine, and amitriptyline (Figure 5b). This
353 result is consistent with that reported by Stevens-Garmon et al. (2011) and Wijekoon
354 Wijekoon et al. (2015) who attributed the notable accumulation of these four compounds onto
355 anaerobic digesters to their moderate hydrophobicity, modest biological persistence, and
356 negative charge. Moreover, the elevated bioreactor salinity could decrease their
357 biodegradation (indicated by the decreased removal by AnMBR, Figure 4b) and thus
358 increased their residue in the digesters (Figure 5b).

359 4 Conclusion

360 Results reported here show that elevated bioreactor salinity negatively affected the
361 performance of AnMBR for wastewater treatment. Both bulk organic removal (indicated by
362 TOC and COD) and biogas/methane production decreased as the bioreactor salinity increased
363 to above 10 g/L NaCl. Of the 33 TrOCs investigated here, the high salinity reduced the
364 removal of most hydrophilic compounds, but insignificantly affected the removal of
365 hydrophobic ones by AnMBR. Moreover, slight impacts on TrOC residues in the sludge were
366 observed with salinity increase. These results suggest that pre-treatment of saline wastewater
367 may be required to ensure the effectiveness and sustainability of AnMBR treatment.

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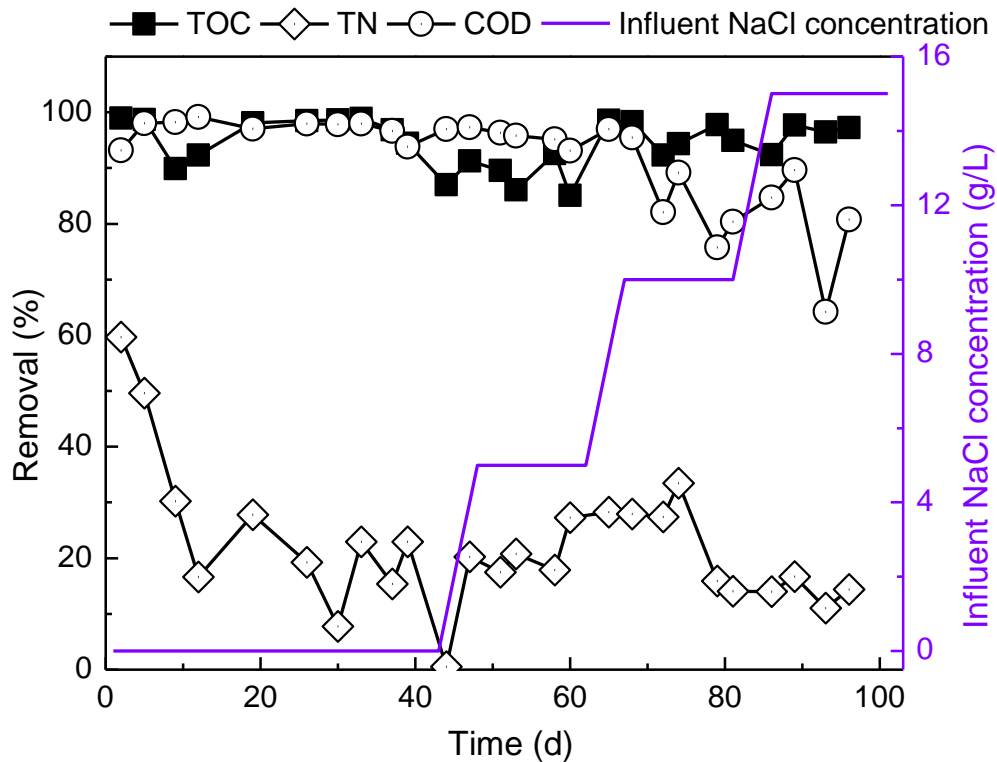
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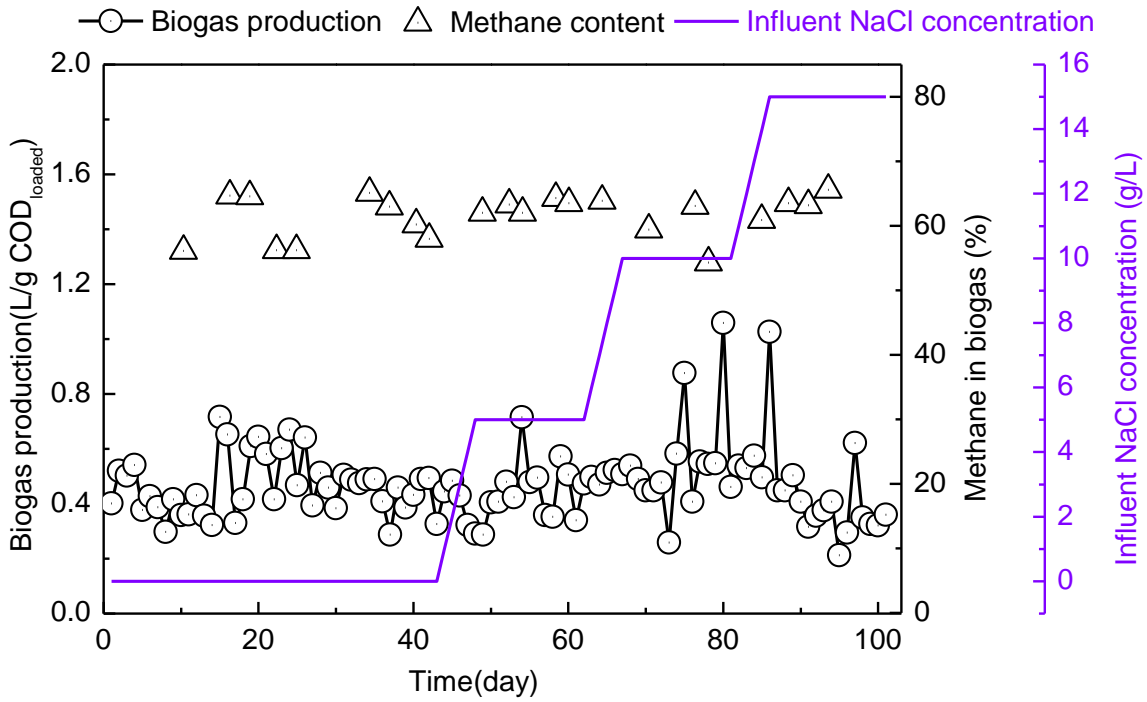
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476 **LIST OF FIGURES**



477

478 **Figure 1:** Effects of salinity build-up in the bioreactor on the removal of bulk organic matter
 479 (i.e. TOC, TN, and COD) by AnMBR. Salinity build-up in the bioreactor was simulated by
 480 increasing the feed NaCl concentration from 0 to 15 g/L. Experimental conditions: initial
 481 MLSS = 16 g/L; HRT = 5 d; mixed liquor pH = 7 ± 0.1 (adjusted by sodium acetate);
 482 temperature = 35 ± 1 °C.

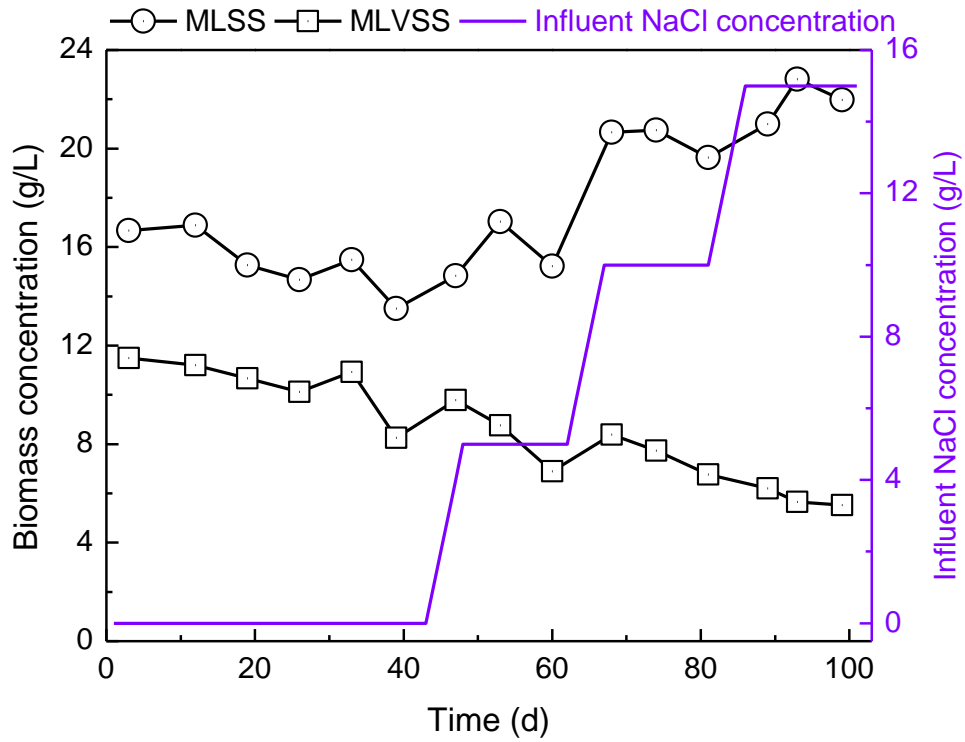


483

484 **Figure 2:** Effect of salinity build-up in the bioreactor on biogas production and its methane
 485 content during AnMBR operation. Salinity build-up in the bioreactor was simulated by
 486 increasing the feed NaCl concentration from 0 to 15 g/L. Experimental conditions are as
 487 described in the caption of Figure 1.

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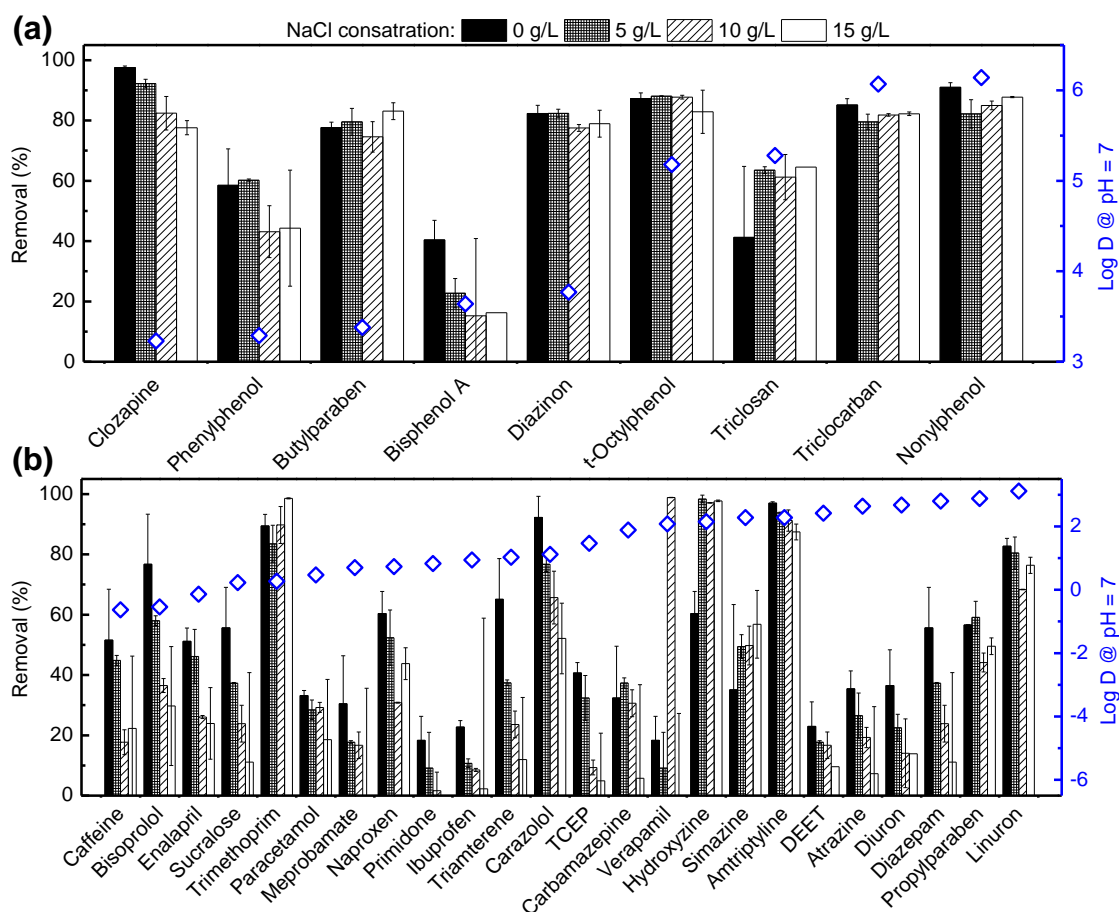
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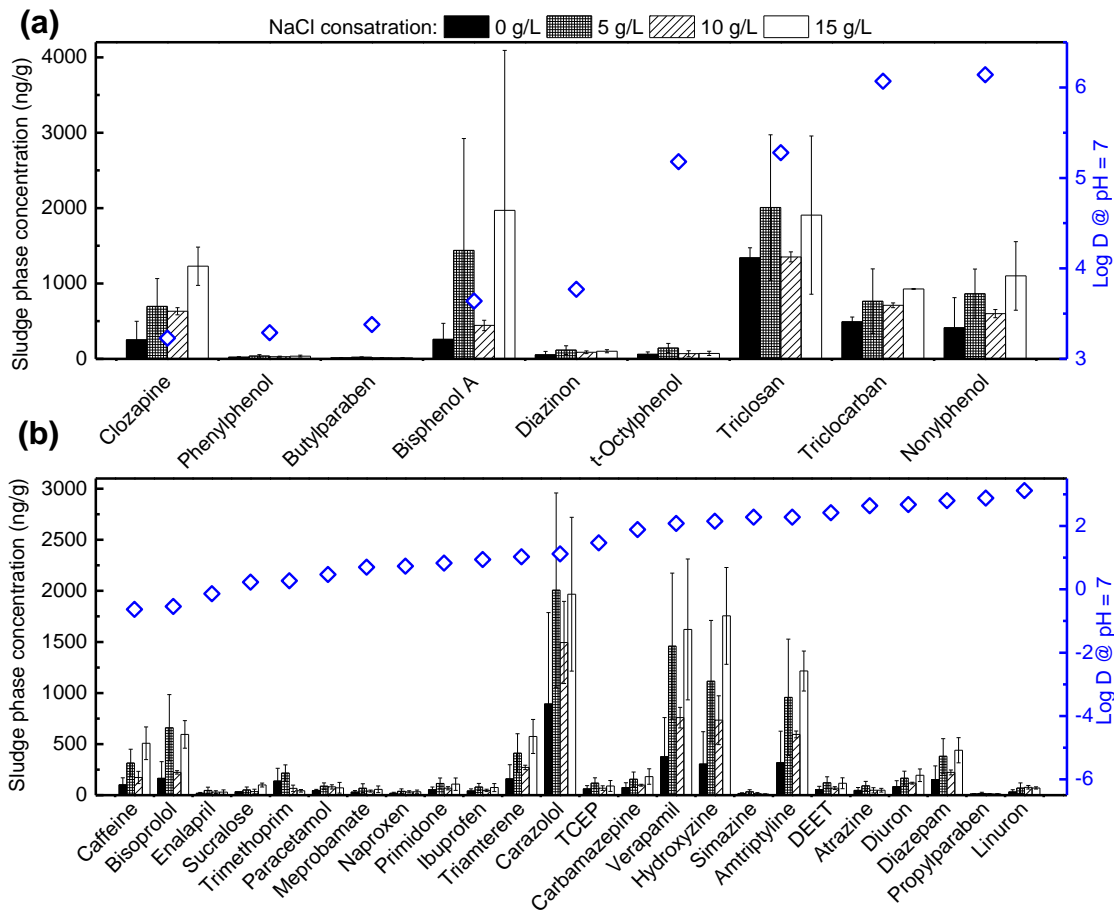
491 **Figure 3:** Effect of salinity build-up in the bioreactor on biomass concentration during
 492 AnMBR operation. Salinity build-up in the bioreactor was simulated by increasing the feed
 493 NaCl concentration from 0 to 15 g/L. Experimental conditions are as described in the caption
 494 of Figure 1.

495



496

497 **Figure 4:** Effects of salinity build-up in the bioreactor on the removal of TrOCs by AnMBR
 498 treatment. The 33 TrOCs investigated could be grouped into hydrophobic (Log D > 3.2 at pH
 499 7) and hydrophilic (Log D < 3.2 at pH 7). Salinity build-up in the bioreactor was simulated
 500 by gradually increasing the feed NaCl concentration from 0 to 15 g/L. To allow microbial
 501 acclimatization to the salinity stress, the influent salt salinity was maintained at 5, 10, and 15
 502 g/L NaCl for two weeks. Error bars represent the standard deviation of two measurements
 503 (once per week) at each salinity condition.



504

505 **Figure 5:** Effect of salinity build-up in the bioreactor on TrOC accumulation in the sludge
 506 during AnMBR operation. Salinity build-up in the bioreactor was simulated by gradually
 507 increasing the feed NaCl concentration from 0 to 15 g/L. To allow microbial acclimatization
 508 to the salinity stress, the influent salt salinity was maintained at 5, 10, and 15 g/L NaCl for
 509 two weeks. Error bars represent the standard deviation of two measurements (once per week)
 510 at each salinity condition.

511 **Effects of salinity build-up on the performance of an anaerobic**
 512 **membrane bioreactor regarding basic water quality parameters**
 513 **and removal of trace organic contaminants**

514

515 *SUPPLEMENTARY DATA*

516 *Submitted to Bioresource Technology*

517 May 2016

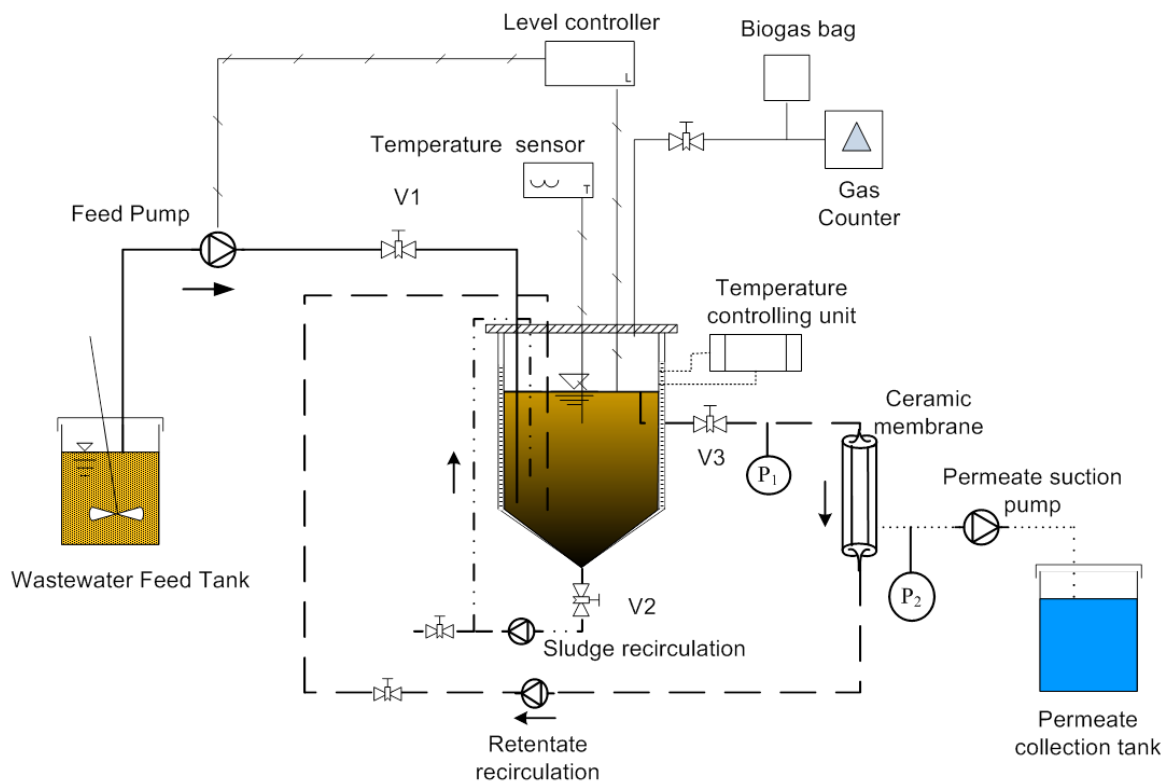
518 Xiaoye Song ^a, James McDonald ^b, William E. Price ^c, Stuart J. Khan ^b, Faisal I Hai ^a, Hao H.
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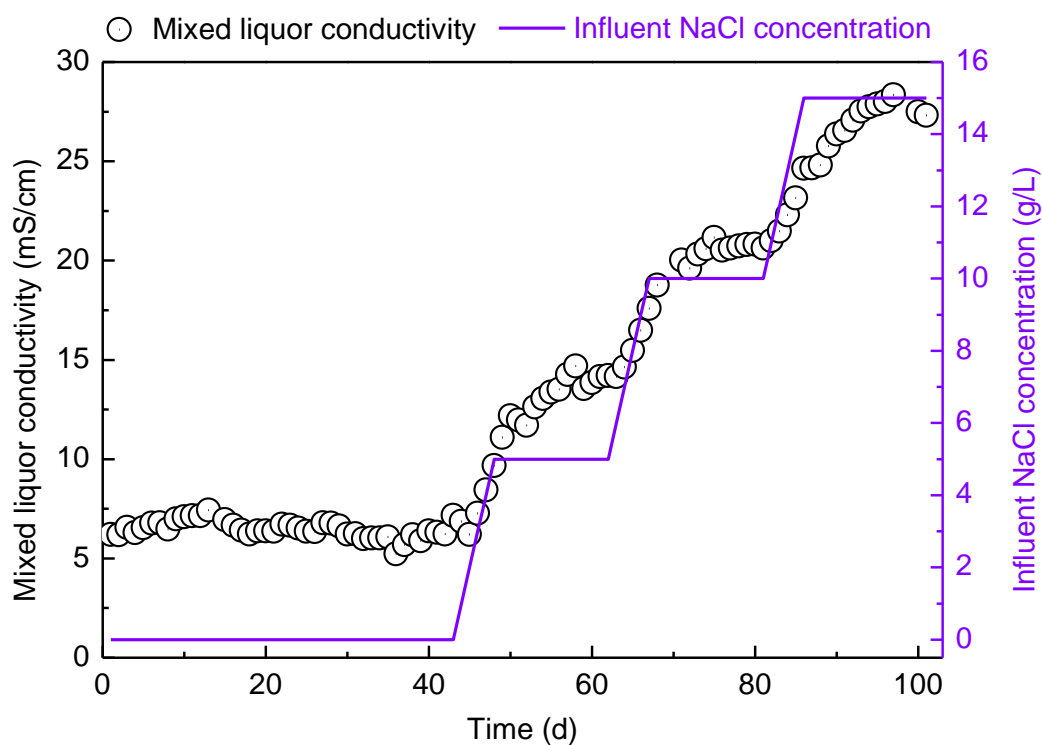
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528

529 **Figure S1:** Schematic diagram of the anaerobic membrane bioreactor (AnMBR) system.

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530

531 **Figure S2:** Increase in the mixed liquor electrical conductivity induced by an increase in the
 532 influent NaCl concentration.

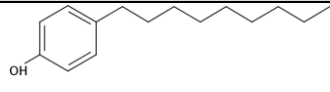
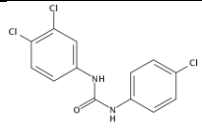
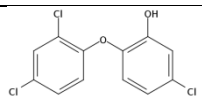
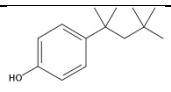
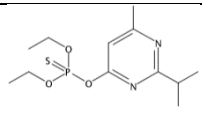
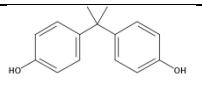
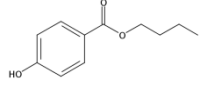
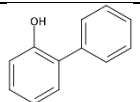
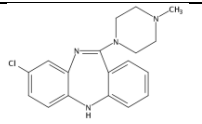
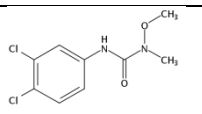
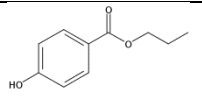
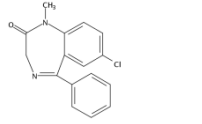
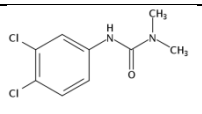
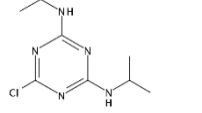
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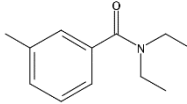
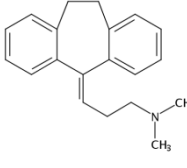
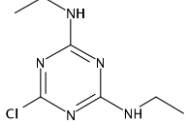
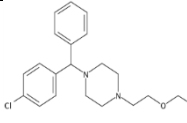
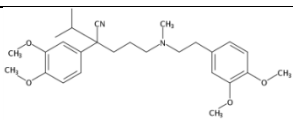
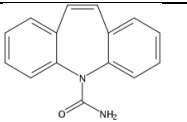
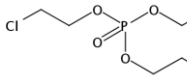
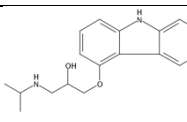
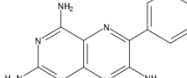
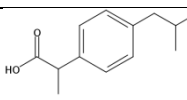
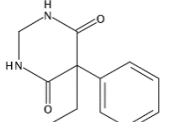
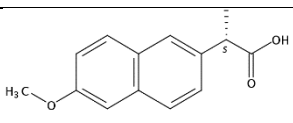
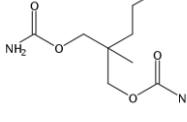
534 **Table S1:** Composition of the synthetic wastewater fed to AnMBR.

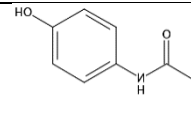
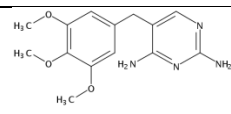
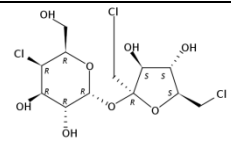
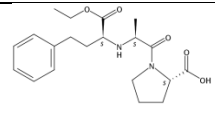
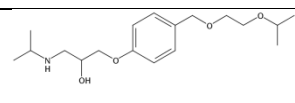
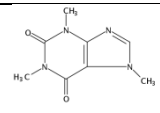
Chemicals	Chemical formula	concentration (mg/L)
Glucose	$C_6H_{12}O_6$	4000
Peptone	-	750
Potassium dihydrogen phosphate	KH_2PO_4	175
Magnesium chloride	$MgCl_2$	175
Sodium acetate	CH_3COONa	2250
Urea	$CO(NH_2)_2$	175
Ferrous chloride	$FeCl_2 \cdot 4H_2O$	45
Nickel chloride	$NiCl_2 \cdot 6H_2O$	10
Cobalt chloride	$CoCl_2 \cdot 6H_2O$	6
Ammonium molybdate	$(NH_4)_6Mo_7O_{24} \cdot 4H_2O$	4

535

536 **Table S1:** Physicochemical properties of the selected trace organic contaminants.

Compounds	Chemical formula	Log D at pH = 7	MW (g/mol)	Chemical structure
Nonylphenol	C ₁₅ H ₂₄ O	6.14	220.35	
Triclocarban	C ₁₃ H ₉ Cl ₃ N ₂ O	6.07	315.58	
Triclosan	C ₁₂ H ₇ Cl ₃ O ₂	5.28	289.54	
t-Octylphenol	C ₁₄ H ₂₂ O	5.18	206.32	
Diazinon	C ₁₂ H ₂₁ N ₂ O ₃ PS	3.77	304.35	
Bisphenol A	C ₁₅ H ₁₆ O ₂	3.64	228.29	
Butylparaben	C ₁₁ H ₁₄ O ₃	3.38	194.23	
Phenylphenol	C ₁₂ H ₁₀ O	3.29	170.21	
Clozapine	C ₁₈ H ₁₉ ClN ₄	3.23	326.82	
Linuron	C ₉ H ₁₀ Cl ₂ N ₂ O ₂	3.12	249.09	
Propylparaben	C ₁₀ H ₁₂ O ₃	2.88	180.20	
Diazepam	C ₁₆ H ₁₃ ClN ₂ O	2.8	284.74	
Diuron	C ₉ H ₁₀ Cl ₂ N ₂ O	2.68	233.09	
Atrazine	C ₈ H ₁₄ ClN ₅	2.64	215.68	

DEET	$C_{12}H_{17}NO$	2.42	191.27	
Amtriptyline	$C_{20}H_{23}N$	2.28	277.403	
Simazine	$C_7H_{12}ClN_5$	2.28	201.66	
Hydroxyzine	$C_{21}H_{27}ClN_2O_2$	2.15	374.90	
Verapamil	$C_{27}H_{38}N_2O_4$	2.08	454.60	
Carbamazepine	$C_{15}H_{12}N_2O$	1.89	236.27	
TCEP	$C_6H_{12}Cl_3O_4 P$	1.47	285.49	
Carazolol	$C_{18}H_{22}N_2O_2$	1.12	298.38	
Triamterene	$C_{12}H_{11}N_7$	1.03	253.26	
Ibuprofen	$C_{13}H_{18}O_2$	0.94	206.28	
Primidone	$C_{12}H_{14}N_2O_2$	0.83	218.25	
Naproxen	$C_{14}H_{14}O_3$	0.73	230.26	
Meprobamate	$C_9H_{18}N_2O_4$	0.70	218.25	

Paracetamol	$C_8H_9NO_2$	0.47	151.16	
Trimethoprim	$C_{14}H_{18}N_4O_3$	0.27	290.32	
Sucralose	$C_{12}H_{19}Cl_3O_8$	0.23	397.63	
Enalapril	$C_{20}H_{28}N_2O_5$	-0.14	376.45	
Bisoprolol	$C_{18}H_{31}NO_4$	-0.54	325.44	
Caffeine	$C_8H_{10}N_4O_2$	-0.63	194.19	

537 Source: SciFinder Scholar (ACS) database.

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