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New Method for the Fabrication of Al-Stabilized Fe/MgB₂ Wires

A. V. Shcherbakov, A. V. Pan, S. X. Dou, and E. W. Collings

Abstract—To employ Al stabilizer on Fe sheathed MgB₂ wire the hot aluminizing technique was used. The present technique allows us to form MgB₂ superconductor by an *in-situ* reaction technique and apply Al stabilizer simultaneously, because the processing temperature of MgB₂ is similar to the melting temperature of Al. Two types of Fe sheathed MgB₂ wires—"reacted" and "green" (unreacted)—were immersed in an Al bath to produce a layer of Al stabilizer on the surface of the superconductor. The magnetic and transport $J_c(B_a)$ curves for both "reacted" and "green" Al/Fe sheathed wires have been measured at 10 and 20 K and showed the same trend. The concept of Al stabilized Fe/MgB₂ coil fabrication is discussed.

Index Terms—Aluminum stabilized superconductor, fabrication method, hot aluminizing, *in-situ* Fe/MgB₂.

I. INTRODUCTION

 \square INCE the discovery of superconductivity in MgB₂ [1], this material has been produced in different forms—bulk, tapes and wires. For practical application of superconductors, wires and tapes with high critical current are necessary. Considerable work has been done to investigate and optimize the properties of MgB2 wires and tapes with different sheath materials such as Ag, Cu, Monel, Ni, Nb, Stainless Steel and Fe (see for example [2]-[10] and references therein). Chemical compatibility with MgB₂, high hardness, low specific weight, and the low price of Fe make it the most suitable sheath material for fabrication of long length MgB2 wires and tapes. For instance, a critical current density as high as 5×10^5 A/cm² at 20 K and 0 T has been achieved in Fe sheathed MgB2 wire [11]. However, Fe creates a low thermal conductivity barrier for heat sinking to the cryogen bath or cryocooler. This can result in thermomagnetic instability [12] and trigger the transition from the superconducting to the normal state in superconducting coils at high currents due to avalanche-like heat release.

It is well known that one of the most important goals for practical large current applications of the wires/coils is good thermal stabilization. To minimize thermomagnetic instability in superconducting MgB₂ wires sheathed in iron, a normal

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metal with high electrical and thermal conductivity, such as Cu or Al, can be used for stabilization. It has been long recognized that high purity aluminum offers several potential advantages in the stabilization of superconductors. Specifically, aluminum is commercially available with a residual resistance ratio (RRR) in excess of 2500, compared to a maximum of approximately 200 for copper. The magnetoresistance of aluminum also saturates rather quickly, so that the specific resistivity of RRR 2500 aluminum is approximately 1/20 that of RRR 200 copper at 12 T. Aluminum also offers well recognized weight and radiation transparency advantages [13], [14]. Attempts to employ Al stabilizer with LTS superconductors such as NbTi and Nb₃Sn were successful, and these results can be found in the literature (see for example [15]-[18] and references therein). In case of MgB₂, however, direct employment of Al for fabrication of Al-sheathed MgB2 wire by the in-situ process will result in serious damage or even loss of superconductivity in case of a reaction between the Al and the unreacted Mg +2B powder during the sintering process [20], [21]. On the other hand, the $J_c(B_a)$ performance of Al/MgB₂ ex-situ wires will not be high enough, due to the low mechanical strength of the Al (see for example [22] and references therein). Different approaches to the production of Al-stabilized/sheathed MgB₂ wires/tapes with reasonably high $J_c(B_a)$ performance can be found elsewhere [22], [23].

To overcome the problems in the fabrication of Al-stabilized MgB₂ wires that were mentioned above, we have employed the hot aluminizing technique [24]. The present technique allows us to apply Al stabilizer to a completely fabricated and optimized Fe/MgB₂ composite. In this case, Fe plays the role of a barrier preventing diffusion of Al into MgB₂. In addition, the high toughness of Fe enables us to produce a reasonably high-density MgB₂ superconducting core. Furthermore, the hot aluminizing technique can be modified to the simultaneous process of fabrication of Al stabilized wires/coils and in-situ MgB2 formation because the processing temperature of in-situ MgB₂ formation (650 °C) is similar to melting temperature of Al (660 °C) [25]. This process is very attractive for the construction of MgB_2 superconducting magnets. In this paper, we report the first results on simultaneous fabrication of *in-situ* Fe/MgB₂ wire and coating this Fe/MgB2 wire with an Al-layer for stabilization purposes.

II. EXPERIMENTAL PROCEDURE

Fe-sheathed MgB₂ wire has been fabricated using the *in-situ* process and the standard PIT method. An Fe tube with an outer diameter (OD) of 10 mm and a wall thickness of 1 mm was filled with a mixture of pure Mg (99%) and amorphous B (99%) with

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the stoichiometry of MgB₂. The composite was drawn to an OD of 1.4 mm. Several short samples were sintered in a tube furnace at 750 °C for 30 min in flowing high purity argon to prevent oxidation. The resultant Fe/MgB₂ with no Al is referred to below as the reference sample. After completing the heat treatment process, the surface of the reacted Fe/MgB2 wires was plated with Al using the hot aluminizing technique. Short samples of reacted Fe/MgB2 wire were immersed for 3 min in a molten Al bath kept at a temperature of 750 °C under flowing high purity argon. After coating, the sample had an OD of 1.32 ± 0.05 mm. These samples are denoted as the "reacted" Al/Fe/MgB2 in the text. To investigate the possibility of a simultaneous process for Al plating and *in-situ* MgB₂ formation, short samples of unreacted Fe/MgB2 wire were immersed for 5 to 30 min in a molten Al bath kept at a temperature of 670 or 770 °C under flowing high purity argon. These samples are denoted as the "green" Al/Fe/MgB₂ in the text.

The phase investigation was performed with the help of an x-ray diffractometer with CuK α radiation. The magnetization loops for the Fe and Al/Fe sheathed MgB₂ wires were measured at 10 and 20 K using a Physical Property Measurement System (PPMS, Quantum Design) with a magnetic field applied perpendicular to the wire long axis. As there is a large sample size effect on the magnetic J_c for MgB₂ fabricated by the reaction *in-situ* process [26], all of the samples for measurement were shaped to the same length of 3.4 mm for comparison. The magnetic J_c was derived from the width of the magnetization loop using Bean's model. The transport J_c was measured with the four-probe method using a pulsed current source. This technique is described in details in [27]. The voltage contacts were attached directly onto the Al surface with Sn-9%Zn solder.

III. RESULTS AND DISCUSSION

Fig. 1 shows XRD patterns for the core of the "green" Al/Fe/MgB₂ samples after immersing unreacted Fe/MgB₂ wire in an Al bath kept at a temperature of 670 °C for 5, 15 and 30 min and at a temperature of 770 °C for 5 and 10 min. As can be seen, at lower temperature (670 °C) the reaction between Mg and B could be achieved after 30 min. Only a small peak of unreacted magnesium could be found. For the higher temperature Al bath (770 °C), the *in-situ* reaction of MgB₂ formation was fully completed in 10 min. A small peak of MgO could be found in all the samples studied. It is to be noted that there is no evidence of any reaction between Al and MgB₂ after the aluminizing routine, indicating the good properties of Fe as a barrier.

SEM investigation of the Al/Fe/MgB₂ wire cross section showed a continuous coating of the Fe sheathed MgB₂ wire with ~ 10 μ m thin Al layer (Fig. 2 and Fig. 3). A diffusion layer of Al₃Fe alloy had a thickness of about 9 μ m at the Al/Fe interface. This diffusion alloy layer indicates good connectivity between the two materials (Fe and Al). It is worth noting that as a result of the hot aluminizing procedure the thickness of the Fe barrier was slightly reduced compared to the Fe/MgB₂ sample due to the reaction between Fe and Al. Increasing the Al bath temperature or time of exposure in hot Al resulted in a further reduction of the Fe barrier thickness (Fig. 3). This is beneficial in case of wire stabilization. As was discussed above, Fe has



Fig. 1. X-ray diffraction patterns for the "green" MgB_2/Fe wire samples immersed in an Al bath at 670 °C for 5, 15 and 30 min and at 770 °C for 5 and 10 min.



Fig. 2. SEM images of the "reacted" Al/Fe/MgB₂.



Fig. 3. SEM images of the "green" Fe/MgB₂ wire sample immersed for 10 min in an Al bath kept at 770 $^{\circ}$ C.

undesirably low thermal conductivity. On the other hand, the Fe barrier should prevent the diffusion of Al into the MgB₂ core that may cause severe damage to the superconductor. The formation of the diffusion Al₃Fe alloy may set limits to the desirable Fe barrier thickness and sintering parameters for the hot aluminizing process.

Fig. 4 shows the magnetic J_c values for the reference Fe/MgB₂, as well as for the "reacted" and "green" Al/Fe/MgB₂ wires with the magnetic field applied perpendicular to the wire long axis. For MgB₂ with the sole Fe-sheath and "reacted" Al/Fe/MgB₂ wire samples magnetic J_c values are similar over the entire range of temperatures and fields measured, although the former is slightly higher than the latter. It is worth noting that the Fe-sheathed MgB₂ wire exhibited more strongly pronounced flux-jump behavior in the low field region (0–2 T) compared to the Al-stabilized MgB₂ wire at 10 K. On the other hand, similar magnetic $J_c(B_a)$ performance indicates that the hot aluminizing process did not affect the properties of the MgB₂ superconducting core. This is a significant advantage of



Fig. 4. Magnetic $J_c(B_a)$ curves at 10 K and 20 K for the reference Fe/MgB₂, and for the "reacted" and "green" Al/Fe/MgB₂ wires. The "green" Al/Fe/MgB₂ wire was reacted in an molten Al bath kept at 670 °C for 30 min.

the aluminizing process, which turns out to be a non-destructive approach to thermomagnetic stabilization.

As can be seen from Fig. 4, Al-coating of the Fe-sheathed MgB_2 wires results in better thermomagnetic stabilization at 10 K. Further improvement of stabilization at temperatures below 10 K can be achieved by increasing the Al to superconductor ratio, decreasing the thickness of the Fe barrier, and reduction of the Fe/MgB₂ wire diameter.

It is to be noted that "green" Al/Fe/MgB₂ wire showed a similar trend in the magnetic $J_c(B_a)$ curve to the reference Fe/MgB₂ and the "reacted" Al/Fe/MgB₂ wire, although the $J_c(B_a)$ values are slightly lower (Fig. 4). At this stage, J_c values of 3.6×10^4 and 2.3×10^4 A/cm² have been reached at 20 K and 3 T for the "reacted" and "green" Al/Fe/MgB₂ wires, respectively. These results are in agreement with those for Fe/MgB₂ wires reported by other groups [6], [10].

The transport $J_c(B_a)$ values of the reference Fe/MgB₂, and the "reacted" and "green" Al/Fe/MgB₂ wires are presented in Fig. 5. The $J_c(B_a)$ curves nearly coincide for Fe-sheathed MgB₂ and "reacted" Al/Fe-sheathed MgB₂ wires at B_a > 4 T at 10 K, as well as at B_a > 3.3 T at 20 K. The transport J_c values of "reacted" Al/Fe/MgB₂ wire are in agreement with previous results on Fe/MgB₂ [5], [7], [28] results, included for comparison in Fig. 5. As can be seen, at 20 K and 3 T the J_c values reached 3.7 × 10⁴, 3.6 × 10⁴ and 2.2 × 10⁴ A/cm² for sole Fe/MgB₂, and the "reacted" and "green" Al/Fe/MgB₂ wires, respectively.

It is expected that the application of the hot aluminizing technique can be further improved by optimization of the melt-processing temperature and dipping time, Fe sheath thickness, and other parameters. Moreover, the aluminizing technique, which is demonstrated in this work for Fe/MgB₂ wires, can be further extended to fabrication of Al-stabilized Fe/MgB₂ coils. We expect that a coil with a wound Fe-sheathed MgB₂ "green" wire, can be heated up to the MgB₂ reaction temperature. While the reaction is complete or in process, the spool can be immersed in a molten Al bath and cooled down. It is obvious



Fig. 5. Transport $J_c(B_a)$ of sole Fe/MgB₂, and the "reacted" and "green" Al/Fe-MgB₂ wires. The "green" Al/Fe/MgB₂ wire was reacted in an molten Al bath kept at 670 °C for 30 min.

that this dual process of *in-situ* MgB₂ formation and coating of Fe/MgB₂ wire/coil by Al will simplify the coil fabrication and markedly improve the cost feasibility of the MgB₂ conductor applications. In addition, Al/Fe/MgB₂ conductors mainly made up of low-density components will be advantageous for airborne, aerospace, and other applications where weight is an important issue.

IV. CONCLUSION

In summary, we have demonstrated a new approach for simultaneous (i) fabrication of *in-situ* Fe/MgB₂ wires and (ii) coating with the Al stabilizer. The transport critical current density of thus-fabricated Al/Fe/MgB₂ wire achieved the level of 2.2×10^4 A/cm² at 20 K and 3.5 T, which is only slightly lower than $J_c = 3.7 \times 10^4$ A/cm² of the reference Fe/MgB₂ sample. The magnetic measurements show that the Al layer stabilizes the Fe/MgB₂ wires in the lower field region at measured temperature range. This method can be easily extended to the production of coils, resulting in reduction of fabrication time and cost of MgB₂ conductor products due to eliminating one processing stage. The light weight of most elements in the Al/Fe/MgB₂ system pave the way for possible use of this superconductor in various applications where weight is a crucial parameter.

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