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Publication Details

This article was originally published as: Humphrey, TE & Linke, H, Reversible Thermoelectric Nanomaterials, Physical Review Letters, 2005, 94(9), 096601. Copyright 2005 American Physical Society. The original journal can be found [here](#).

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Reversible Thermoelectric Nanomaterials

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(Received 20 July 2004; published 9 March 2005)

Irreversible effects in thermoelectric materials limit their efficiency and economy for applications in power generation and refrigeration. While electron transport is unavoidably irreversible in bulk materials, here we derive conditions under which reversible diffusive electron transport can be achieved in nanostructured thermoelectric materials. We provide a fundamental thermodynamic explanation for why the optimum density of states in a thermoelectric material is a delta function and for why inhomogeneous doping and segmentation improve the thermoelectric figure of merit.

DOI: 10.1103/PhysRevLett.94.096601

PACS numbers: 72.20.Pa, 05.70.-a, 73.63.Kv, 85.35.Be

The textbook examples for heat engines are *cyclic* engines, such as the well-known Carnot, Otto, and Brayton cycles, which may be used to model steam turbines or gasoline engines [1]. Cyclic heat engines transfer heat between at least two reservoirs via a working gas that moves through a number of quasiequilibrium states. Reversibility may be achieved when the cycle progresses infinitely slowly and heat transfer is isothermal [1].

Less well known is a distinct type, which we here denote as *particle exchange* (PE) heat engines [2]. These transfer heat between at least two reservoirs via the continuous exchange of particles in a finite energy range in the presence of a field against which work is done. Reversibility is achieved when particles are transmitted only at the energy where the occupation of states in the reservoirs is equal [3–5]; heat transfer is then isentropic but nonisothermal [2]. Most PE heat engines studied previously have been discrete, with particles moving elastically between two reservoirs only, and include the three-level amplifier [2,3], solar cells and light emitting diodes [2,4], and ballistic electron heat engines [2,5,6].

Nanostructured thermoelectric materials can have dramatically higher efficiencies than their bulk counterparts [7–11]. While the precise physical effects responsible for this improvement in certain materials are not well understood, theoretically it has been shown that the higher interface density in thermoelectric nanomaterials can reduce losses due to phonon heat conductivity [12] and also that a sharply peaked electronic density of states (DOS) (as may, in principle, be achieved in low-dimensional materials [11,13,14]) is optimal for high efficiency [15].

Here we model thermoelectric nanomaterials from a thermodynamic point of view as *continuous* PE heat engines in which electrons in a relatively narrow energy band move diffusively through a material with a continuous spatial variation in temperature and electrochemical potential. We derive conditions under which thermoelectric nanomaterials can operate reversibly, so challenging a long held view that thermoelectric devices are inherently

irreversible heat engines [16]. Our results have significant potential for practical application in the design and optimization of new thermoelectric nanomaterials with very high limiting efficiencies.

The efficiency of any heat engine is bounded above by the Carnot limit, which can be achieved only in systems infinitesimally close to an equilibrium state. An electronic system in equilibrium is characterized by a spatially invariant occupation of states given by the Fermi-Dirac (FD) distribution function,

$$f_{\text{FD}} = [\exp(s^{+1}(\mathbf{r})/k) + 1]^{-1}, \quad (1)$$

where $s^{+1}(E, \mathbf{r}) = [E - \mu(\mathbf{r})]/T(\mathbf{r})$ corresponds to the entropy change in the system if one electron with energy E is added to the system at the spatial coordinate \mathbf{r} . A spatially invariant probability of occupation of available electronic states can be achieved in three ways, two of which are well known, while the third, a continuous form of the result for discrete PE heat engines, is pointed out here.

First, a state of global equilibrium is attained when the electrochemical potential $\mu(\mathbf{r})$ and the temperature $T(\mathbf{r})$ are both constant as a function of \mathbf{r} . This corresponds to the textbook definition of an equilibrated electronic system.

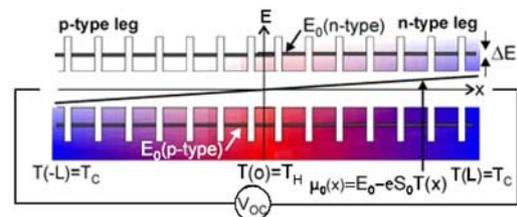


FIG. 1 (color online). Band structure of a thermoelectric quantum dot superlattice (QDSL) or a superlattice nanowire (SLNW) as used in our numerical model. Narrow minibands with width ΔE for electrons (holes) are located at energy E_0 ($-E_0$). The doping level varies across the material to ensure that $\mu(x)$ satisfies Eq. (3). Note that for simplicity, band bending near the metallic contacts has not been considered.

Second, equilibrium can be approached when the only available electronic states in the system are at very high energies (for instance, in an intrinsic semiconductor with band gap, $E_g \rightarrow \infty$) where occupation tends to a spatially constant value of zero, irrespective of finite gradients in $\mu(\mathbf{r})$ and $T(\mathbf{r})$.

Here, we identify a third way in which equilibrium can be approached in continuous electronic systems, which we denote “energy-specific equilibrium” [17]. We consider a material in which (i) the DOS for electrons is a delta function at E_0 and in which (ii) $\mu(\mathbf{r})$ and $T(\mathbf{r})$ vary across the system in such a way that $s^{+1}(E_0, \mathbf{r})$ is spatially invariant. This means that the population of electron states at the specific energy E_0 is the same throughout the material [see Figs. 1 and 2(c)]. Under these conditions the entire electronic system is in equilibrium, in spite of the thermal and potential gradients.

In the following we show how energy-specific equilibrium can be implemented in order to achieve reversible electron transport. We note that the rate of entropy production per unit volume, \dot{s} , in the material due to the movement of electrons in response to temperature and electrochemical potential gradients is [1]

$$\dot{s} = \nabla \cdot \left(\frac{1}{T} \right) (J_q + \mu J_n) - \nabla \cdot \left(\frac{\mu}{T} \right) J_n, \quad (2)$$

where J_n is the number flux of electrons in the direction of decreasing temperature and J_q the heat flux due to electrons, and where μ , T , and J_q depend on \mathbf{r} . In the limit that only electrons with energy E_0 are transmitted, the heat current in the material is given by $J_q(\mathbf{r}) = J_n[E_0 - \mu(\mathbf{r})]$. To find the conditions under which electrons move

through the material without increasing the entropy of the system (reversible transport) we set $\dot{s}(E_0) = 0$ to obtain a differential equation which may be solved by integration with respect to \mathbf{r} to find $E_0 = \mu(\mathbf{r}) + \Omega T(\mathbf{r})$. Here Ω is an integration constant, evaluated using the boundary condition $\mu_C - \Omega T_C = \mu_H - \Omega T_H$, where the subscripts C and H refer to the cold and hot extremes of the system, respectively. At open circuit, this yields $\Omega = -eV_{OC}/\Delta T$, where $eV_{OC} = \mu_C - \mu_H$, $\Delta T = (T_H - T_C)$, and where ΔT and μ_C (or alternatively μ_H) are freely chosen parameters. We then arrive at our main result, namely, an expression for a spatially varying chemical potential that ensures that electron transport at E_0 is reversible at open circuit:

$$\mu_0(\mathbf{r}) = E_0 - eV_{OC}[T(\mathbf{r})/\Delta T]. \quad (3)$$

(The subscript 0 refers to the state of energy-specific equilibrium.) To clarify the physics involved in Eq. (3) we note that

$$S_0 \equiv V_{OC}/\Delta T = [E_0 - \mu_0(\mathbf{r})]/eT(\mathbf{r}) \quad (4)$$

is the thermopower (Seebeck coefficient) corresponding to energy-specific equilibrium, which can be physically interpreted as the entropy carried by 1 A of current [1] [that is, $eS_0 = s^{+1}(E_0)$]. As S_0 is spatially invariant, there is no entropy increase when an electron with energy E_0 moves through the material and, therefore, no thermodynamically spontaneous direction for current to flow, despite the finite thermal and electrical potential gradients, confirming that the electronic system is in equilibrium.

In practice, reversible electron transport can be approached by (i) creating a nanostructured material, such as a QDSL or a SLNW [9,18] with a DOS for mobile electrons that is sharply peaked at one energy E_0 [which could, in principle, depend on \mathbf{r} so that $E_0 = E_0(\mathbf{r})$ [19]] and that is (ii) inhomogeneously doped such that Eq. (3) is fulfilled (Fig. 1) [19]. Note that a given material can be optimized only for one specific temperature gradient.

A material with a delta function DOS doped according to Eq. (3) will have an “electronic” efficiency (efficiency in the absence of phonon heat leaks and other nonidealities) approaching the Carnot limit. To show this, we note that the heat flux withdrawn from the hot end of the system by electrons is $|J_q(T_H)| = (E_0 - \mu_H)|J_n|$ [Eq. (3)]. At open circuit the power is $P \equiv eV_{OC}|J_n| = eS_0(T_H - T_C)|J_n|$, giving an electronic efficiency $\eta_{PG} \equiv P/J_q(T_H) = (1 - T_C/T_H)$, which is the Carnot limit for power generation. Similarly, if the system is operated in reverse ($J_n \rightarrow -J_n$) as a refrigerator, then the coefficient of performance can be shown to be equal to the Carnot limit, $\eta_R \equiv J_q(T_C)/P = [T_C/(T_H - T_C)]$.

S_0 is, in fact, the theoretical upper bound upon the Seebeck coefficient $S(\mathbf{r})$ for a particular value of $[E_0 - \mu(\mathbf{r})]$ and $T(\mathbf{r})$. To show this, we set $\mu(\mathbf{r}) = \mu_H$ and $T(\mathbf{r}) = T_H$ and assume that a distance $\delta\mathbf{r}$ from \mathbf{r} the potential in-

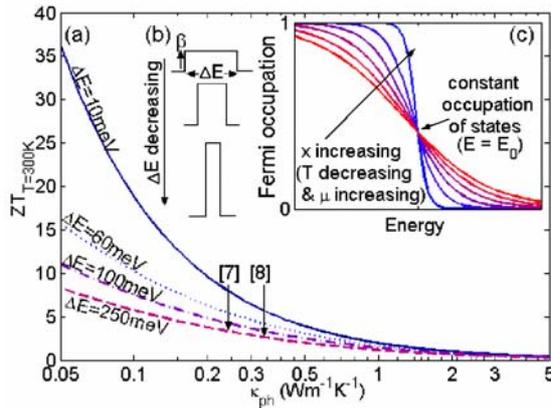


FIG. 2 (color online). (a) ZT at $T_C = 300$ K as a function of κ_{ph} for different ΔE as indicated. Experimental values are marked [7] and [8] for κ_{ph} obtained at 300 K in the nanomaterials described in the corresponding references. (b) Schematic of the variation of the magnitude of β with decreasing ΔE resulting from the scaling operation described in the caption to Fig. 3. (c) Fermi occupation function at several places along the x axis in the material shown in Fig. 1.

crease is $\delta\mu$ and the temperature decrease is δT . Then using Eq. (4) we find that $eS_0\delta T = (E_0 - \mu_H)\delta T/T_H$. This means that $S = S_0$ when the Carnot fraction, $\delta T/T_H$, of heat removed by each electron with energy E from the hot end of the small section of material, $(E_0 - \mu_H)$, is converted to useful work $eS_0\delta T = \delta\mu$. S_0 therefore represents a theoretical maximum, as $S > S_0$ would imply an efficiency greater than the Carnot limit.

It is important to note that the presence of inelastic scattering in thermoelectric materials does not affect these results. Inelastic scattering processes produce heating via the relaxation of carriers from a nonequilibrium occupation of states to an occupation given by a FD distribution with the appropriate local value of $\mu(\mathbf{r})$ and $T(\mathbf{r})$. Since in our system states at E_0 are occupied with the same probability throughout the material [see Fig. 2(c)], inelastic processes that scatter individual electrons into these states are just as likely as processes that scatter electrons out of these states, so there is no net exchange of energy between free electrons and the lattice due to the movement of electrons with energy E_0 .

If the DOS is finite in some range around E_0 , electrons occupying states at energies above (below) E_0 are, on average, scattered to lower (higher) energy states, depending upon the difference between the local probability of occupation and that of the arriving electrons. This process results in local heating (cooling) of the lattice. Crucially, however, the nearer these electrons are in energy to E_0 , the smaller the heating (cooling) effect, due to the small variation in the occupation of states between adjacent regions of the material.

To quantify the advantage of using a deltalike DOS and inhomogeneous doping according to Eq. (3), we now numerically characterize a nanomaterial in which we vary the DOS from deltalike to bulklike. We assume a finite lattice thermal conductivity ($\kappa_{ph} \neq 0$) and a single miniband of width ΔE as shown in Fig. 1. We use the Boltzmann transport equation under the relaxation-time approximation [20]. Electrical conductivity, σ , thermal conductivity due to electrons, κ_{el} , and S can all be expressed as a function of the integral [20]

$$K_\alpha = \int \beta(E)(E - \mu)^\alpha \left(-\frac{df}{dE}\right) dE, \quad (5)$$

where $\beta(E) = D(E)\tau(E)v(E)^2$, $\tau(E)$ is the electron relaxation time, $v(E)$ is the electron group velocity, $D(E)$ is the DOS, and where $\alpha = 0, 1$, or 2 . Then $\sigma = e^2K_0$, $S = -K_1/(eTK_0)$, $\kappa_{el} = (K_2 - K_1^2/K_0)/T$, and the dimensionless figure of merit $ZT = T\sigma S^2/(\kappa_{el} + \kappa_{ph})$. We use $\kappa_{ph} = 0.33 \text{ W m}^{-1}\text{K}^{-1}$, the value measured by Harman *et al.* for PbSeTe/PbTe QDSL [8]. For simplicity and transparency in the numerical results, we assume that $\beta(E) = \beta$ is constant over the energy range ΔE , a reasonable assumption for the small values of ΔE in which we are primarily interested. To isolate the effect upon ZT of changing ΔE

from effects due to changing the overall *number* of states, we scale the magnitude of β with ΔE such that for all values of ΔE , $\sigma = 5 \times 10^5 \text{ } \Omega^{-1}\text{m}^{-1}$ at $\mu_c = E_0$ [see Fig. 2(b) and the Fig. 3 caption for details]. This choice means that at the values of μ_c for which ZT is optimized at $T_C = 300 \text{ K}$, we obtain $\sigma \approx 6 \times 10^4 \text{ } \Omega^{-1}\text{m}^{-1}$ for $\Delta E \approx 200 \text{ meV}$, the same conductivity as measured in [8] (note that the energy spectrum of the DOS in [8] is not known). In addition, we obtain $\sigma \approx 1 \times 10^5 \text{ } \Omega^{-1}\text{m}^{-1}$ for an optimized ZT at $\Delta E \approx 60 \text{ meV}$, which is similar to the value obtained by numerical modeling in [9] for a PbSe/PbS SLNW with a 60 meV wide miniband.

Figure 3(a) shows S as a function of $(E_0 - \mu_c)$ at the position $x = L$ for different ΔE . As ΔE decreases the system approaches energy-specific equilibrium and S approaches S_0 , the theoretical maximum Seebeck coefficient given by Eq. (4).

Figures 3(b) and 3(c) show σ and κ_{el} , respectively, as a function of $(E_0 - \mu_c)$. Note that although σ at $\mu_c = E_0$ is kept constant with decreasing ΔE , κ_{el} at $\mu_c = E_0$ decreases; that is, $K_1^2/K_0 \rightarrow K_2$ [15]. The physical reason for this is that as the heat carried by an electron is proportional to the difference between its energy and the Fermi energy, materials with narrow DOS ($\Delta E/2$ less than several kT), which ‘‘cut off’’ the high energy end of the Fermi distribution, have low κ_{el} . An important implication of this result is that the Wiedemann-Franz law, which expresses that $\kappa_{el}/\sigma T = (\kappa\pi)^2/3e^2$ [20], and which is often used by experimentalists to calculate κ_{el} from the electrical conductivity [7,8], loses validity in nanomaterials with narrow DOS. Crucially, this result also means that thermoelectric nanomaterials are not subject to the conundrum that limits

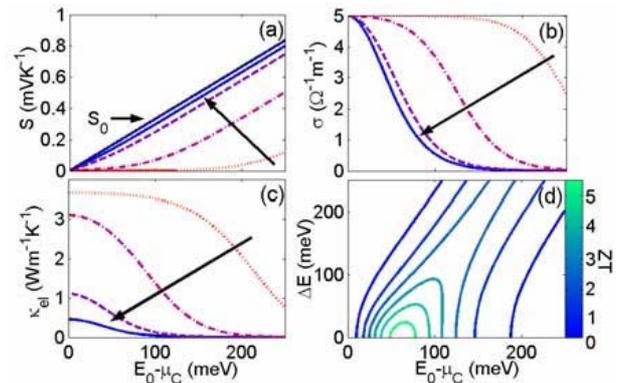


FIG. 3 (color online). Numerically calculated thermoelectric parameters as a function of $(E_0 - \mu_c)$ for $\Delta E = 500, 250, 100$, and 60 meV (arrows indicate decreasing ΔE), for $T(L) = 300 \text{ K}$, and for $\kappa_{ph} = 0.33 \text{ W m}^{-1}\text{K}^{-1}$. For each ΔE we used the value $\beta = 5 \times 10^5/e^2K_0(\Delta E, \mu_c = E_0)$ to calculate K_0, K_1 , and K_2 for $\mu_c \neq E_0$. See the text for further model details. (a) Seebeck coefficient. (b) Electrical conductivity. (c) Electronic thermal conductivity. (d) Contour map of ZT . (a) is antisymmetric and (b)–(d) mirror symmetric with respect to $(E_0 - \mu_c) = 0$ (not shown).

ZT of bulk thermoelectrics: the interrelationship of S , σ , and κ_{el} .

To emphasize this point, we show in Fig. 3(d) that for small ΔE the figure of merit is optimal at values of $(E_0 - \mu_C)$ [21] where S [Fig. 3(a)] and κ_{el} [Fig. 3(c)] are both relatively small. In other words, the best strategy to maximize ZT in nanomaterials is to minimize κ_{el} leaving ZT limited only by finite κ_{ph} . The fact that this is possible without at the same time decreasing σ illustrates a fundamental difference between the thermodynamics of nanostructured and bulk thermoelectric materials.

Using the technique outlined in [22], the difference between keeping $[E_0 - \mu(x)]/T(x)$ constant across a thermoelectric nanomaterial [i.e., $\mu(x) = \mu_0(x)$] via inhomogeneous doping or segmentation and allowing $[E_0 - \mu(x)]/T(x)$ to vary [i.e., $\mu(x) = \mu_0(L/2)$] can be quantified. We find [23] that keeping $[E_0 - \mu(x)]/T(x)$ constant can increase the maximum efficiency by 10% (corresponding to a doubling of ZT at 300 K for $\Delta E = 10$ meV, $T_H = 800$ K, $T_C = 300$ K, and $\kappa_{ph} = 0.5$ W m⁻¹ K⁻¹) and increase the maximum power by up to 60%. The physics behind this improvement is now clear; inhomogeneous doping and segmentation increase ZT as they bring thermoelectric materials closer to a state of energy-specific equilibrium.

Figure 2(a) shows ZT at 300 K as a function of κ_{ph} , demonstrating that while a decrease in κ_{ph} is always beneficial to the figure of merit, a decrease in κ_{ph} combined with a decrease in ΔE results in spectacular increases in ZT . The development of nanomaterials with a κ_{ph} that is 20% lower than in current state-of-the-art materials and optimized, deltalike DOS could result in $ZT = 10$ at $T = 300$ K, well within the range of $ZT > 5$ required for economical adoption of thermoelectric technology for mainstream refrigeration and power generation.

Finally, we note the finite coherence length of electrons, which places a lower limit on the width of DOS peaks resulting from quantum confinement, will also, in principle, limit the efficiency of thermoelectric nanomaterials. However, as all heat engines are in reality operated at finite power, away from maximum efficiency, quantum efficiency limits are not expected to be a practical design issue.

We thank H. J. Goldsmid for helpful comments on the manuscript. T. H. is supported by the ARC and SERDF, and H. L. by NSF (PHY 0239764).

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