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Abstract

Dissociation of superoxide dismutase 1 dimers is enhanced by glutathionylation, although the dissociation constants reported to date are imprecise. We have quantified the discrete dissociation constants for wild-type superoxide dismutase 1 and six naturally occurring sequence variants, in their unmodified and glutathionylated forms, at the ratios expressed. Unmodified superoxide dismutase 1 variants that shared similar dissociation constants with SOD1WT had disproportionately increased dissociation constants when glutathionylated. This defines a key role for glutathionylation in superoxide dismutase 1 associated familial amyotrophic lateral sclerosis.

Keywords

toxic, 1, variants, potentiates, sclerosis, glutathionylation, lateral, amyotrophic, associated, forms, superoxide, dismutase, benign, CMMB

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Glutathionylation potentiates benign superoxide dismutase 1 variants to the toxic forms associated with amyotrophic lateral sclerosis

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Dissociation of superoxide dismutase 1 dimers is enhanced by glutathionylation, although the dissociation constants reported to date are imprecise. We have quantified the discreet dissociation constants for wild-type superoxide dismutase 1 and six naturally occurring sequence variants, in their unmodified and glutathionylated forms, at the ratios expressed. Unmodified superoxide dismutase 1 variants that shared similar dissociation constants with SOD1^{WT} had disproportionately increased dissociation constants when glutathionylated. This defines a key role for glutathionylation in superoxide dismutase 1 associated familial amyotrophic lateral sclerosis.

Approximately one in five cases of familial amyotrophic lateral sclerosis (fALS) is due to naturally occurring mutations in the gene encoding superoxide dismutase 1 (SOD1)¹. Over 150 such mutations have been reported to occur, and these mutations can occur at virtually any residue of the primary sequence. Many of these mutations confer a significant level of instability upon the native dimer, which may ultimately unfold and form insoluble cytoplasmic aggregates - a hallmark of SOD1 associated fALS pathology². Recently, evidence has emerged to suggest that SOD1^{WT} aggregation is also involved in some cases of the considerably more prevalent sporadic ALS³⁻⁵.

The prevailing hypothesis of SOD1 aggregation suggests that dissociation of the native homodimer is a key step, as SOD1 monomers have been identified as a common misfolding intermediate^{6,7}. This mechanism is supported by the extraordinary difference in dissociation constant (K_D) reported for the SOD1^{A4V} dimer (3 μ M)⁸, compared with SOD1^{WT} (2.25 nM)⁹; i.e. the dissociation of SOD1^{A4V} into monomers is approximately 1000 times more facile at concentrations found *in vivo*. Furthermore, chemical cross-linking of the aggregation prone variant, SOD1^{A4V} can prevent its aggregation *in vitro*⁸, while cross-linking of other mutants conferred greater thermostability¹⁰.

A compounding factor in SOD1 dissociation is the post-translational modification (PTM) of Cys111 by the ubiquitous antioxidant, glutathione (GSH), which has been demonstrated as being the dominant PTM in human tissue⁹ (Fig. 1a; Fig. S1a). Similar to many residues in the primary chain of SOD1, Cys111¹¹ is proximal to the dimer interface and, not surprisingly, the introduction of GSH has been proposed to augment dissociation¹¹. We explored this hypothesis for a set of human SOD1 variants with well-defined and broad ranging rates of disease progression; from 1.2 years survival post-onset for SOD1^{A4V}, to approximately 17.6 years for SOD1^{H46R}.

Results

Our initial assay, using mass spectrometry, revealed that all variants were glutathionylated to some extent when produced in *E. coli*, from a low of 3% for SOD1^{V148G} to 36% for SOD1^{A4V} (Fig. 1a). When these data were arranged in order of disease progression, a plausible correlation was evident for SOD1^{A4V}, SOD1^{G93A}, SOD1^{G37R} and SOD1^{H46R}; however substantial glutathionylation of non-pathogenic SOD1^{WT} compared to the aggressive variant SOD1^{V148G}, suggested that this modification alone is not predictive of disease progression. Indeed, while the sequence of these SOD1 variants may determine their susceptibility to glutathionylation, further work is required to confirm this in ALS patients.



To compare the relative dissociation propensity of our SOD1 variants, we performed nano-electrospray-ionization mass spectrometry, with conditions optimized to preserve solution phase structures during ionization and subsequent *in vacuo* detection¹² (Fig. 1b; Fig. S2). Each of the SOD1 variant spectra exhibited peaks arising from two dominant ion populations at ~ 2900 m/z and ~ 3200 m/z , corresponding to the 10^+ and 11^+ ions of their respective dimers. Only minor peaks were observed for the 9^+ and 12^+ charge states, indicating that the dimers had negligible variation in their tertiary and quaternary structures. On closer inspection, a series of three maxima were identified within the 10^+ and 11^+ peaks, arising from dimers composed of; (i) two unmodified SOD1 (*uSOD1*) monomers; (ii) *uSOD1* paired with *gsSOD1*; (iii) two *gsSOD1* monomers. With the exception of SOD1^{A4V}, monomers were significantly less abundant than dimers, but similarly confined to two (7^+ and 8^+) charge states (Fig. 1b & inset). For all variants, the most abundant monomer observed was *gsSOD1* (Fig. 1c), even though *uSOD1* constituted the majority of total protein.

The mixture of SOD1 dimers was further interrogated using tandem mass spectrometry to isolate an ion population exclusive to the dimer (D^{11+}), prior to collision-induced dissociation (CID). At the minimum accelerating voltage (6 V) required to traverse an argon-filled CID region, the spectrum contained a single large peak of isolated D^{11+} ions (Fig. S3a). At an accelerating potential of 20 V, two additional peaks arose at ~ 2700 m/z and 3200 m/z , being due to the dissociation of D^{11+} into charge-complementary M^{5+} and M^{6+} monomers. These peaks increased in intensity at 40 V and 60 V, after which the more asymmetric dissociation pathway via M^{4+} and M^{7+} monomer ions became competitive. This hierarchical gas-phase molecular division, based on charge, was also observed for *gsSOD1* (Fig. S3b). We compared the M^{6+} abundance for *uSOD1* and *gsSOD1* under low (10 V) versus high (70 V) energy CID conditions: at 10 V, *gsSOD1*^{A4V}, *gsSOD1*^{G93A} and *gsSOD1*^{G37R} had a greater proportion of M^{6+} than *uSOD1* indicating that less energy is required to remove *gsSOD1* monomer from the dimer than *uSOD1*. A comparison of the spectra at 70 V versus 10 V shows that *gsSOD1*:*uSOD1* is larger at low energy CID for all seven proteins, which supports the hypothesis that modification by glutathione destabilizes the dimer.

A caveat to the gas-phase analysis of molecular interactions is the absence of bulk water which, to varying degrees, can affect the strength of non-covalent association between proteins¹³. Thus, it was possible that the more facile removal of *gsSOD1* from the gas-phase dimer (Fig. 1b&c) is not necessarily reflective of the aqueous environment found *in vivo*. To address this uncertainty, we prepared dilution series of the SOD1 variants ranging from 20 μ M down to 625 nM, prior to mass analysis using identical instrument conditions for all samples. The spectra acquired at each concentration provided a snapshot of all species present in solution and so, by plotting the proportion of monomer versus concentration (Fig. 2a; Fig. S4), we were able to calculate discrete dissociation constants for (*uSOD1*)₂ and (*gsSOD1*)₂ (Table 1; Fig. 2b). Unlike the homodimers, which dissociate into their respective and unique monomer pairs, products of the mixed dimer (*uSOD1*:*gsSOD1*) contributed to both populations of monomer, and were not so readily estimated. In an effort to account for monomers originating from the mixed dimer, we monitored the decline in (*uSOD1*:*gsSOD1*)¹¹⁺ ion intensity with successive dilution, and subtracted an equivalent quantity from each monomer signal. Although imprecise, we believe this correction attenuated, to some extent, the confounding effects of the mixed dimer species.

All *uSOD1* variants had K_D values greater than *uSOD1*^{WT} (9 ± 1 nM), implying that these single residue substitutions are sufficient to disturb the quaternary structure (Table 1). The most substantial augmentation was observed for *uSOD1*^{A4V} (1.1 ± 0.1 μ M), being approximately 100-fold greater than *uSOD1*^{WT},

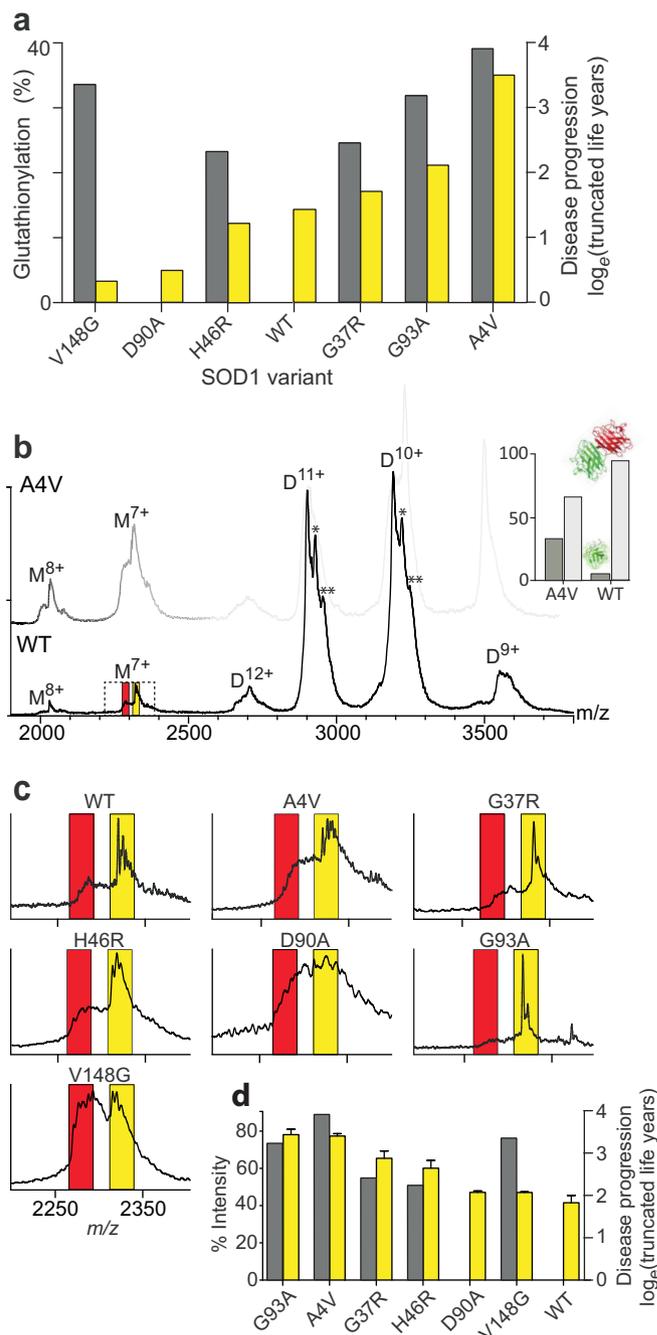


Figure 1 | Mass spectrometry analysis of SOD1 variants. (a) Percentage of recombinantly expressed SOD1 that was glutathionylated (yellow), as determined by the area under the peaks of MS spectra, compared to disease progression (grey), expressed in truncated life years as $\log_e[(75 - (25 + d))]$, where d equals the reported average disease duration, assumed life expectancy is 75 years and age at onset of disease is set to 50 years. (b) Mass spectrum of native SOD1^{WT} (main) and SOD1^{A4V} (faint) with species charge states labeled (D = dimer, M = monomer). Additional maxima within the peaks are the result of glutathionylation of one or both subunits to give a mixed (*) or *gsSOD1*(**) dimer. Inset shows the ratio of dimer to monomer for the two proteins, derived from charge state peak areas in the spectrum. (c) Expansion of the M^{7+} region showing peaks corresponding to *uSOD1* (red) and *gsSOD1* (yellow). (d) The percentage of *gsSOD1* monomer (yellow) compared to disease progression (grey) for all SOD1 variants. The *gsSOD1* measurements were performed in triplicate and error bars represent standard error of the mean.

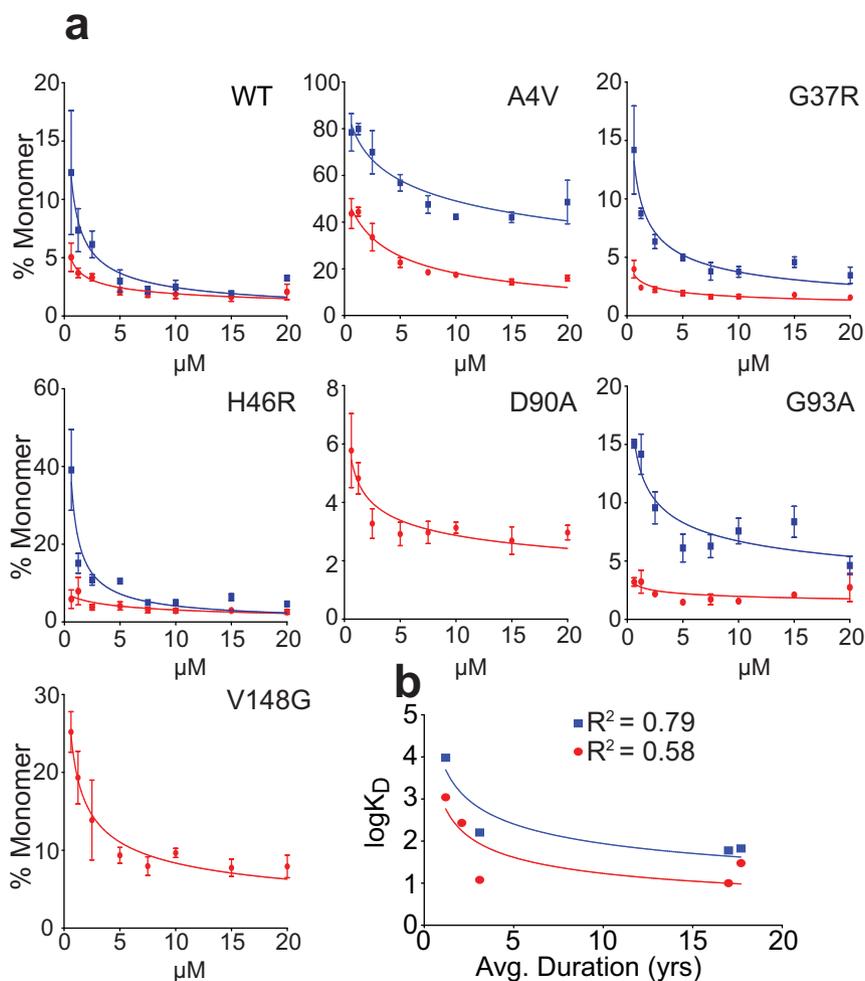


Figure 2 | SOD1 monomer abundance increases with dilution. (a) of SOD1 variants showing changes in the proportion of monomeric μ SOD1 (red) and gsSOD1 (blue) as a function of protein concentration: (i) SOD^{WT}, (ii) SOD1^{A4V}, (iii) SOD1^{G37R}, (iv) SOD1^{H46R}, (v) SOD1^{D90A}, (vi) SOD1^{G93A}, (vii) SOD1^{V148G}. Analyses were performed in triplicate at each concentration and error bars represent standard deviation about the mean. (b) Relationship between disease duration and K_D for μ SOD1 (red) and gsSOD1 (blue).

whereas μ SOD1^{G93A} (12 ± 1 nM) had a K_D only marginally higher than μ SOD1^{WT}. Significantly, the K_D value of each gsSOD1 species was higher than the corresponding μ SOD1, with the largest (13-fold) increase observed for gsSOD1^{G93A}, followed by gsSOD1^{A4V} (9-fold), gsSOD1^{G37R} (6-fold) and gsSOD1^{WT} (4-fold) (Table 1). There was insufficient gsSOD1^{V148G} and gsSOD1^{D90A} to accurately calculate their K_D values.

Discussion

Glutathionylation of SOD1 expressed in *E. coli*, has previously been reported to occur at each of the four cysteine residues (Cys6, Cys57, Cys111, and Cys146) as detected by mass spectrometry¹⁴. Contrastingly, our analysis of recombinant SOD1^{WT} demonstrated that no more than one glutathione was bound per subunit, and this was the case for all SOD1 variants in this study (Fig. S1). This discrepancy

stems from the protocol used by Furukawa and coworkers¹⁴, which directed expression of SOD1 into inclusion bodies or soluble apo-proteins followed by refolding and metal enrichment after extraction from bacterial cells¹⁵. In the present work, SOD1 was co-expressed with the yeast-copper-chaperone for SOD1 (yCCS) in the presence of copper and zinc, with subsequent purification of soluble SOD1 from the bacterial cytosol¹⁶. The yCCS facilitates folding of the nascent chain, and formation of the disulfide bond between Cys57 and Cys146¹⁴. This disulfide effectively locks the β -barrel structure, rendering these Cys residues inaccessible to larger molecules; *ipso facto*, unlikely sites of glutathionylation. Furthermore, it has been demonstrated that Cys6 is buried inside the β -barrel of catalytically active SOD1, i.e., inaccessible to aqueous reductants¹⁷. We assessed the dismutase activity of our variants using in-gel zymography, which yielded positive results for all but the metal-binding site mutant,

Table 1 | SOD1 dissociation constants as determined by native mass spectrometry. The dissociation constants are means \pm standard error from 3 distinct data sets with each containing 5 data points (see supplementary figure 4)

Variant	WT	A4V	G37R	H46R	D90A	G93A	V148G
SOD1 K_D (nM)	9 ± 1	1100 ± 100	10 ± 1	30 ± 4	25 ± 4	12 ± 1	271 ± 40
GS-SOD1 K_D (nM)	34 ± 5	9600 ± 900	60 ± 9	97 ± 16	-	160 ± 32	-



SOD1^{H46R} (Fig. S1b). Therefore, although we did not explicitly demonstrate Cys111 to be the site of our observed glutathionylation, it is inferable that the only cysteine residue sufficiently exposed and therefore substantially modified by glutathione was Cys111.

It is well documented that patients suffering from SOD1-associated fALS have variable survival times, dependent upon the particular amino acid mutation¹⁸. Considering the role played by dimer dissociation, as an entrée to aggregation, we scrutinized our data for evidence of a link between patient survival and K_D (Fig. 2b). For $uSOD1$, the K_D values were poorly associated with disease duration ($R^2 = 0.58$), primarily due to the outlier characteristics of $uSOD1^{G93A}$. In contrast, K_D values for $gsSOD1$ showed a much stronger correlation ($R^2 = 0.79$), indicating that glutathionylation may be responsible for the disparity between perceived SOD1 stability and observed patient survival. Consider the G93A mutation. This is reported as having little effect upon dimer dissociation¹⁹, and our results for the K_D of $uSOD1^{G93A}$ confirm this (Fig. 3a); yet the survival time of patients carrying this mutation is, on average, a mere 2.2 years. Our ability to distinguish between, and calculate separate K_D values for $uSOD1$ and $gsSOD1$ resolves this apparent inconsistency - the K_D of $gsSOD1^{G93A}$ is 13 times higher than that of $uSOD1$ (Fig. 3b; Table 1) - highlighting the significance of this methodology.

Our data can be used to reframe the original hypothesis in more specific terms: i.e., it is not only (the extent of) SOD1 glutathionylation that influences the rate of disease progression, but also the K_D of the glutathionylated species. Glutathionylation, however, is not a requirement for pathogenicity, as sequence modification alone of

SOD1^{V148G} causes a 30-fold increase in K_D over SOD1^{WT}. Similarly, our extracted K_D value for $uSOD1^{A4V}$ reflects the considerable instability of this non-glutathionylated species. These mutations are located deep in the inter-subunit cleft, where small molecular volume changes are able to disrupt the dimer interface leading to more facile dissociation²⁰ (Fig. S5).

In contrast to Val148 and Ala4, Gly37 and Gly93 are spatially remote to the dimer interface, being situated in the $\beta 3/\beta 4$ Greek key loop and $\beta 5/\beta 6$ inter-strand loop respectively (Fig. S5). It is notable that these loops are proximal to each other and the Leu38 cork, while Cys111 is within the $\beta 6/\beta 7$ Greek key loop at the opposite end of the β -barrel. Prior to glutathionylation, the individual mutations G37R and G93A impart negligible effect upon the K_D of SOD1; i.e., *in isolation, these mutations are not pathologically destabilizing* (Table 1). Rather, it is the synergy of glutathionylation and point mutation at the β -barrel ‘poles’ that accounts for the significantly higher K_D values reported in this study, and presumably, the malignance of these variants *in vivo*. It stands to reason then, that glutathionylation of SOD1 alters the K_D of a multitude of individual variants in a unique, allosteric-like fashion, and that it is the combination of $K_D^{(uSOD1)}$ and $K_D^{(gsSOD1)}$ (Fig. 3c) which impact the pathogenicity of the protein. Therefore, we propose an elaboration to the contemporary notion in which SOD1 dissociation is the primary underlying cause of SOD1 fALS.

Specifically, dissociation of the SOD1 dimer is a biophysical consequence of either, (i) a potent destabilizing mutation such as A4V, or (ii) glutathionylation of a weak or non-destabilizing mutation

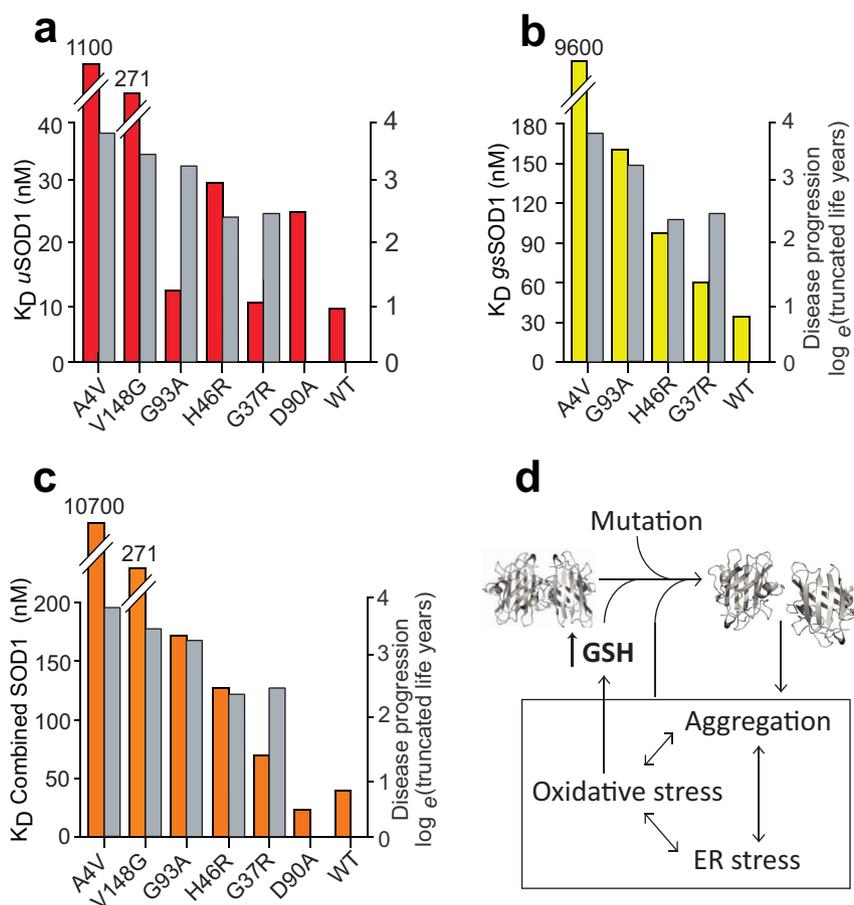


Figure 3 | The profound effect of glutathionylation on SOD1 dissociation constants. Histograms comparing K_D values of (a) $uSOD1$, (b) $gsSOD1$, and (c) $uSOD1 + gsSOD1$, to disease progression (grey). (d) A scheme of the proposed factors affecting SOD1 dimer dissociation. Briefly, dimer dissociation occurs due to mutation, glutathionylation, and cellular stresses. The increased monomer population is more likely to unfold and increase cellular stress. Glutathione (GSH) levels in the cell are elevated during periods of oxidative stress, thereby increasing the chances of SOD1 becoming glutathionylated.



such as G93A. Consequently, there is a twofold increase in the abundance of monomers, which are more susceptible to misfold into toxic conformations compared to the dimer^{6,7} (Fig. 3d). If the cellular clearance rate of misfolded SOD1 is inadequate, increasingly frequent hydrophobic interactions ultimately give way to aggregation, forming larger amorphous structures that stress the cellular machinery²¹. Many neurodegenerative diseases, including ALS are also associated with oxidative stress²², which in turn can lead to an up-regulation of glutathione²³. Indeed, an increase in GSH has been observed in ALS patient CSF²⁴. Analysis of human tissue from ALS patients, and in cell experiments, are currently under way to further investigate the role of glutathionylation in SOD1 associated fALS.

Methods

Materials. All solutions were prepared using Milli-Q ultra-purified water using the Millipore purification system (Massachusetts, USA). All materials were of analytical grade. Tryptone, Tris-base, DTT, β -mercaptoethanol, bromophenol blue, brilliant blue, ammonium persulphate, SDS, ampicillin sodium salt, and IPTG were from Amresco (Ohio, USA). NaCl, ammonium sulphate, agar, acetonitrile, Tris-HCl, tetramethylethylenediamine, ammonium acetate and RNase were from Sigma-Aldrich (Missouri, USA). DNase was from Roche Diagnostics (NSW, Australia). Carbenicillin was from Carbenicillin Direct (UK). Copper sulphate pentahydrate was from Ace Chemical Company (SA, Australia). Zinc sulphate heptahydrate was from Hopkin and Williams LTD (Birmingham, UK). Formic acid and acetic acid were from Univar (NSW, Australia). Methanol was from Ajax Finechem (NSW, Australia). Gold coated borosilicate electrospray capillaries were made in-house.

Expression and purification of recombinant SOD1. Plasmids encoding SOD1^{WT}, SOD1^{A4V}, SOD1^{G37R}, SOD1^{H46R}, SOD1^{D90A}, SOD1^{G93A}, and SOD1^{V148G} were a gift from the Oliveberg Group (Sweden). Protein expression and purification were performed according to¹⁶. Briefly, the SOD1 plasmids were transformed into chemically competent BL21* *E. coli* using heat shock. Following transformation, the protein was expressed in the presence of copper and zinc ions with the addition of IPTG. The expressed SOD1 was purified via size exclusion chromatography (Hiload 16/60 Superdex 75 PG, GE USA) and anion exchange chromatography (Hiscreen Capto-Q, GE USA).

Sample preparation for mass spectrometry. SOD1 samples were desalted and buffer exchanged into 200 mM NH₄OAc (pH 7), using gel filtration chromatography (Superdex 200 10/300 GL, GE USA). Briefly, SOD1 samples were concentrated to 20 mg/ml and 100 μ l was loaded onto the column at a flowrate of 0.4 mg/ml. The concentration of SOD1 was determined at 265 nm using a UV/Vis spectrophotometer, and a molar extinction coefficient of 18 700 M⁻¹cm⁻¹²⁵.

Mass spectrometry. SOD1 samples in 200 mM NH₄OAc were denatured using formic acid and acetonitrile with the final concentration of each being 10% and 40% respectively. Mass analyses were performed on a Q-ToF ULTIMA mass spectrometer (Waters, UK) in positive ion mode with nano-electrospray source. Instrument parameters included: capillary 1.5 kV, sample cone 200 V, desolvation gas flow 180 L/h. Native protein analyses were performed in 200 mM NH₄OAc on a SYNAPT G2 HDMS (Waters, UK) in positive ion mode. Instrument parameters were similar to Ruotolo et al. (2008)²⁶ with the exceptions: capillary voltage 1.5 kV, sample cone 30 V, cone gas 70 L/h, trap collision voltage 6 V, ion-transfer stage pressure 4.15 e⁻¹ mbar, ToF analyzer pressure 2.38 e⁻⁶ mbar. All SOD1 variants were analyzed at a concentration of 20 μ M. For MS/MS experiments, the trap collision energy was increased from 10 to 100 V in increments of 10 V with 30 scans at each increment. The dilution series mass spectra were acquired at concentrations of 20, 15, 10, 7.5, 5, 2.5, 1.25 and 0.625 μ M with the following instrument parameters: capillary voltage 1.52 kV, sample cone 137 V, cone gas 78 L/h, trap collision voltage 16 V, transfer CE 8 V, ion transfer stage pressure 3.72 e⁻⁴ mbar, ToF analyser pressure 9.01 e⁻⁷ mbar, backing pressure 4.0 e⁹ mbar. All mass spectra in this study were externally calibrated using a solution of cesium iodide (10 mg/ml in water) and were processed using Masslynx 4.1 software (Waters, UK).

Dissociation constants. Dissociation constants were derived from the mass spectrometry data using the method of Rose et al., (2011)²⁷. Briefly, the intensity, as determined by the area under the peak, of all peaks relating to SOD1 monomers (M_i) and dimers (D_i) were summed to give total signal intensity (I_{TS}):

$$I_M + I_D = I_{TS} \quad (1)$$

The proportion of monomer signal (P_M) was determined by dividing the intensity of the monomeric species (I_M) by the total signal intensity (I_{TS}):

$$I_M / I_{TS} = P_M \quad (2)$$

The concentration of monomer at equilibrium ([M]_{eq}) was determined by multiplying the proportion of monomer signal (P_M) by the total protein concentration in μ M as determined by UV/Vis spectroscopy ([P₀):

$$P_M \times [P_0] = [M]_{eq} \quad (3)$$

The concentration of dimer at equilibrium ([D]_{eq}) is defined as:

$$([P_0] - [M]_{eq}) / 2 = [D]_{eq} \quad (4)$$

The K_D was derived from the gradient of the plot of [D]_{eq} against [M]_{eq}² (fitting performed with Prism 5.0, GraphPad Software, Inc.) In each case at least 5 data points were plotted with repeat experiments being performed if the line of best fit gave least-squares-regression, R² < 0.75.

Statistical analysis. Statistical analyses were carried out using Prism 5.0 software (Graphpad). All mass analyses were performed in triplicate (n = 3) with error bars representing the standard error of the mean. The dissociation plots (Figure 2) were fitted with a Log (Gaussian) function. The K_D plots (Fig. S4) were fitted using linear regression.

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Author contributions

JAA and JJY conceptualized, designed and supervised the study. LM performed experiments, analyzed data, and wrote the initial manuscript draft. JAA revised and finalized the manuscript for submission.

Additional information

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